

**Literature Compilation and Data Synthesis  
for Atmospheric Deposition  
to the Tampa Bay Watershed**

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## Literature Compilation and Data Synthesis for Atmospheric Deposition to the Tampa Bay Watershed

### Background

Remedial efforts to improve impacted surface waters have predominantly focused on the removal and/or improvement of point source discharges. Comparatively recently, efforts have been expanded to address non-point source loadings. Atmospheric contributions of materials to estuaries and their watersheds have been largely ignored. Selected case studies (Paerl, 1985; Fanning, 1989; Winchester and Fu, 1992, Hicks, *et al.*, 1992), however, indicate that atmospheric deposition of nitrogen can comprise substantial portions of an estuarine nutrient budget. While first order estimates of atmospheric deposition can and have been prepared from existing long term data sets (Zarbock, *et al.*, 1993), most long term data have been collected in relatively rural locations, in an attempt to describe regional patterns of deposition. The watershed of Tampa Bay, however, has a significant urban component, particularly near the Bay, with a large number of stationary sources as well as mobile emissions of nitrogen oxides and metals. There is concern that loadings determined from rural areas may significantly underestimate actual loadings. In addition, there is very little information on atmospheric deposition of toxic metals in Florida, and again the major studies underway at this time are, for the most part, located in relatively un-urbanized areas.

As a result, the Tampa Bay National Estuary Program has funded a program to refine deposition estimates. The project is an outgrowth of determining current and benchmark nutrient loadings for Tampa Bay and establishing pollutant reduction goals necessary to support and restore the Bay's essential natural resources. Pollutants scheduled for monitoring are nitrogen, phosphorus, copper, lead, and zinc.

Nitrogen is of interest since the majority of Tampa Bay appears to be nitrogen limited with chlorophyll content and total nitrogen concentrations strongly linked (Johansson, 1991). In addition, the organically bound nitrogen in atmospheric deposition is poorly characterized and will be analyzed as well. The phosphorus content of rainfall is typically low, only a small percentage is thought to be transported via the atmosphere (Duce, 1986), and national monitoring networks use the presence of phosphorus to indicate a contaminated sample (Dr. Jay Gibson, personal communication). The industry of phosphate ore mining and processing in the Tampa Bay region, however, may contribute particulate phosphorus species in excess of levels determined elsewhere, and previous Tampa-specific wet deposition data (Zarbock, *et al.*, 1993; Noel, *et al.*, 1987) could account for substantial phosphorus deposition direct to the Bay. Fugitive emissions from bulk loading facilities in Hillsborough Bay have also been the subject of some discussion (Cardinale and Dunn, 1991; Orlando, 1993).

Additionally, sediment studies of the estuarine portion of Tampa Bay have identified some areas anthropogenically enriched in toxic metals (Schropp, 1990; Alexander, *et al.*, 1993; Long, *et al.*, 1991), particularly cadmium, mercury, lead, silver, and zinc. Some fish in the Hillsborough River exceed quality criteria for mercury content (FDEP, unpublished data; Hand and Friedemann, 1990). Metals have long been identified with urban runoff, and rainfall data indicate that substantial portions of metal loading can be attributed to atmospheric deposition. There is no national network, however, for monitoring the wet deposition of metals (Vermette,

*et al.*, 1992). Copper (Cu), lead (Pb), and zinc (Zn) were selected based on the importance of anthropogenic inputs, toxicity, and mobility. Mercury, while of critical interest to the TBNEP program, was deleted from the analytical suite due to the ongoing Florida Department of Environmental Protection (FDEP) Florida Atmospheric Mercury Study (FAMS) and the technical rigors and expense of obtaining and processing contamination-free samples.

The following compilation of literature pertinent to Tampa Bay will discuss sources, emissions, available ambient air quality data, precipitation quality, and the reported ranges of wet and dry deposition rates. Due to the relative scarcity of synoptic quantitative data for all regions of the state, data from many differing time periods will be presented. These data should be viewed with the precaution that precipitation chemistry and resultant deposition rates are notoriously variable and require long-term records for valid spatial or temporal comparisons. The text on sources and emissions will focus on nitrogen and phosphorus species, as sources of the toxic metals to be addressed by the study were to be summarized in the Toxic Contamination Action Plans, under production concurrently with this document. Data on atmospheric deposition of metals are even scarcer, and so pertinent data from other regions of the county are presented. Data have been transformed as necessary from the originally reported ionic concentrations and loads (i.e. mg/l NO<sub>3</sub><sup>-</sup>) to elemental concentrations and loadings (i.e. as N). Other units have been similarly converted to the S .I. system.

## Introduction

Atmospheric deposition to an estuary can occur through either direct or indirect mechanisms. Direct inputs include both wet and dry deposition directly to the surface of the waterbody. Wet deposition is the mechanism by which particulate, aerosol, or gaseous materials are dissolved in or physically removed from the atmosphere by rainfall. Dry deposition includes the gravitational settling of larger particles, the impaction and retention of aerosol particles onto a surface, and the absorption of gaseous materials onto solid or liquid surfaces. The individual chemical species of a pollutant class have primary modes of deposition which are a function of the physical state (solid, aerosol, or gaseous), dominant particle size, chemical characteristics, receiving surface characteristics, and micrometeorology near the receiving surface.

For indirect deposition to an estuary, both wet and dry deposition occur on the land, waterbodies, structures, and vegetation surfaces of the watershed. The portion of materials which is subsequently transferred to the tributaries and eventually to the estuary through rainfall, runoff, and tributary flow is termed the indirect input. The transmission factor, or the percent which ultimately leaves the watershed, will vary by pollutant species, being controlled by the physical absorption onto soils and vegetation, as well as biogenic uptake and recycling within the watershed.

Precipitation quality is linked to ambient air quality, and thus emissions, with generally higher atmospheric fluxes observed in areas with higher emissions and ambient air concentrations (Luecken, *et al.*, 1992). Seasonal variations in rainfall patterns, emissions, and prevailing winds often produce seasonal variations in wet deposition loadings (Luecken, *et al.*, 1992). Dry deposition of larger particulates is driven by gravitational settling and typically occurs quite close

to emissions sources (Scudlark, *et al.*, 1992). The dry deposition of gaseous and aerosol materials is more a function of boundary layer turbulence and receptor surface, with dry deposition increasing with increased turbulence, increased surface area (leaf area index), and surface moisture (dew or waterbody surface) (Sherwood, 1991).

### Global and Regional Sources

Sources of materials to the atmosphere can be either “natural” or “anthropogenic” with natural sources including biogenic processes, outgassing from soils or volcanic activity, forest fires, resuspension of crustal material, and aerosol formation from surface waters. Anthropogenic sources typically include energy production and other fossil fuel combustion (industrial activities and transportation), waste incineration, mining, and related manufacturing.

Anthropogenic sources of nitrogen oxides are almost exclusively from combustion of fossil fuels, either stationary sources such as utilities and industrial boilers, or mobile sources such as transportation. Under high temperature combustion, nitrogen and oxygen dissociate and combine to primarily form nitrogen dioxide (NO<sub>2</sub>) and nitrogen oxide (NO), collectively known as NO<sub>x</sub>. The higher the flame temperature, the higher the NO<sub>x</sub> production. In addition, nitrogenous compounds in fuel react with oxygen during combustion to produce NO<sub>x</sub> (Kaplan, 1991). Of the various fossil fuels, coal generates the highest NO<sub>x</sub> emissions per unit of energy produced (Neuffer, 1985). For municipal solid waste incinerators as a group, fuel NO<sub>x</sub> may comprise up to 80 % of the total NO<sub>x</sub> generated (Radian, 1991). Photochemical reactions of NO<sub>x</sub> with hydrocarbons subsequently result in ozone, while reactions with water vapor result in nitric acid (HNO<sub>3</sub>), a component of acid rain.

Minor sources of anthropogenic NO<sub>3</sub><sup>-</sup> in rainfall might include fertilizer applications or soil entrainment during agricultural operation or land clearing. Natural sources of NO<sub>3</sub><sup>-</sup> in rainfall include oxidation of NO<sub>2</sub> in biological decay and NO<sub>3</sub><sup>-</sup> produced from lightning and N<sub>2</sub>. Atmospheric ammonia (NH<sub>3</sub> and NH<sub>4</sub><sup>+</sup>), aside from industrial releases (such as may be used in utility NO<sub>x</sub>-emission control systems) or fertilizers, is primarily a biogenic decomposition product (Gleick, 1993). Anthropogenically induced concentrations of these species, therefore, result wherever concentrated animal or human wastes are allowed to decompose, with 75 % of atmospheric concentrations estimated to come from feed lots and fertilizer applications (Placet, *et al.*, 1990 as cited in Rushton, 1993).

Nationwide, NO<sub>x</sub> emissions were relatively stable between 1979-1988 at approximately 18 million metric tons/year with 55 % allocated to fuel combustion and 41% to transportation (USEPA, 1990). Annual mean air quality in 1988 of 116 sites was approximately 25 ppb of NO<sub>x</sub> (USEPA, 1990). Through the reconstruction of historical fuel usage and transportation, NO<sub>x</sub> emissions in the southeast are thought to have increased roughly exponentially since the 1900's (Husar, *et al.*, 1993) with emissions approaching 6.4 million metric tons NO<sub>2</sub> per year by the early 1980's. Uncertainties in this figure could be as high as a factor of 2.

On a national scale, work on the empirical relationships between emissions of NO<sub>x</sub>, within various sized geographic cells and wet deposition of NO<sub>3</sub><sup>-</sup> (Luecken, *et al.*, 1992) observed the

best correlations between rainfall quality and a 720 km radius cell. Implications are that most  $\text{NO}_x$  is transported a considerable distance before removal by rainfall. Such relationships were strongest in the northeastern United States.

Sources of phosphorus to the atmosphere include resuspension of crustal materials, vegetation combustion products, and marine aerosols. Anthropogenic sources in addition to burning are typically thought to include fertilizer applications and handling, and phosphate mining.

Natural sources of trace metals to the atmosphere include forest fires, sea salt aerosols, and soil dust, but globally, anthropogenic sources exceed natural releases by a factor of 28 for lead and 3 for zinc. Copper and mercury anthropogenic emissions are equal to or slightly greater than natural releases (Verry and Vermette, 1991; Nriagu, 1991). Anthropogenic emissions are, in addition, concentrated in the industrialized countries, thus increasing their relative impacts in these regions. In the atmosphere, crustally derived trace metals are typically contained in coarser particles ( $> 2.5$  microns) and deposited near the source, whereas metal oxides vaporized in high temperature combustion form sub-micron sized aerosols (Scudlark, et al., 1992).

Globally, anthropogenic lead is primarily contributed to the atmosphere through combustion of lead gasoline additives (75 %), smelting (14 %), manufacturing and battery construction (5 %), and energy (coal-fired) production (4%) (Nriagu and Pacyna, 1988, as summarized in Nriagu, 1991). Waste incineration accounts for less than 1%. Copper and zinc have similarly large contributions from smelting and refining, 66% and 55 %, respectively, with energy production accounting for 23 % of anthropogenic copper emissions and 13% of zinc (Nriagu and Pacyna, 1988, as summarized in Nriagu, 1991). National trends in both lead emissions and ambient lead concentrations show dramatic tenfold declines since the late 1970's, from 108,700 metric tons/year to 7.6 metric tons/year. From near 90% in 1979, transportation sources of lead now comprise only 34% (USEPA, 1990).

Between one quarter and one half of the anthropogenic mercury emitted is estimated to come from waste incineration (Verry and Vermette, 1991; Windham, in press) with 64 % estimated from energy production (Nriagu and Pacyna, 1988, as summarized in Nriagu, 1991). For mercury, as for other metals and  $\text{NO}_x$ , emitted pollutants depend heavily on the source fuel, with solid waste and coal typically responsible for higher levels of metals emissions per energy unit produced.

Almost all atmospheric mercury exists in the relatively insoluble elemental form. Oxidation is required before mercuric ions can be effectively scavenged by precipitation (Porcella, *et al.*, 1991). Mean atmospheric residence times of about a year imply that global cycling controls most deposition (Porcella, *et al.*, 1991). Oxidized species such as mercuric chloride are more susceptible to removal by precipitation (and relatively easy to remove with emissions controls), but the proportion of the total mercury emitted in this form depends on the chlorine content of the fuel. Reportedly, municipal incinerators emit a larger percentage of mercury as mercuric chloride, with a potential for higher levels of deposition near the point source (Windham, in press). Little quantitative evidence exists, however, of localized increases in deposition near sources of mercuric ions or oxidants (Porcella, *et al.*, 1991).



## Study Area

The watershed to Tampa Bay includes approximately 5895 km<sup>2</sup> located predominantly in three counties; Pinellas, Hillsborough, and Manatee, with additional portions in Pasco, Polk and Sarasota Counties (Figures 1 and 2). Of the watershed, 324 km<sup>2</sup> represent internal drainage, leaving 5571<sup>2</sup> of watershed contributing to Tampa Bay. The water surface of the Bay itself is approximately 958 km<sup>2</sup> (Zarbock, *et al.*, 1993). The airshed, however, the region from which emissions can influence deposition within either the Bay or the watershed, is a potentially much larger region. The exact shape of the airshed cannot be rigidly defined as it varies annually, seasonally, hourly, and instantly with local and regional climatological patterns. The transport of some pollutants can occur over thousands of kilometers, and, coupled with plume dispersion from stationary or area sources, makes absolute source attribution a difficult task.

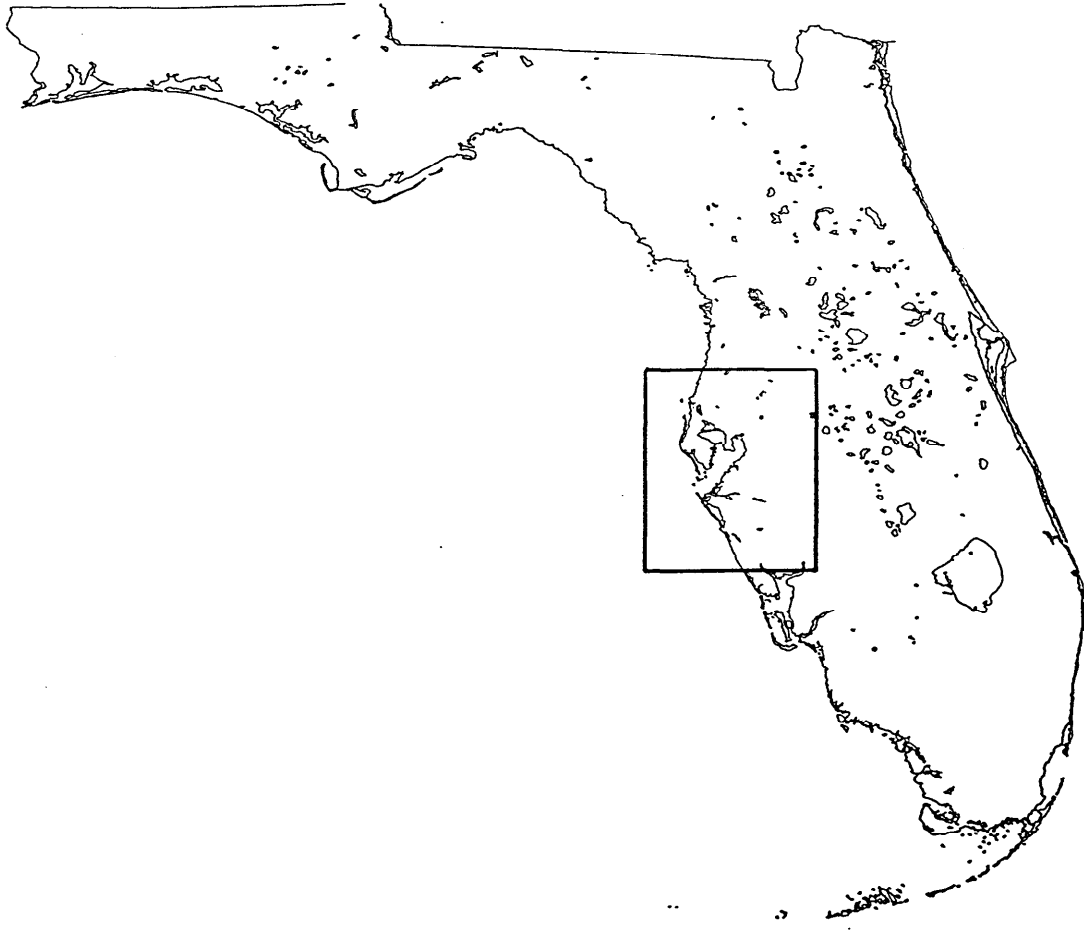


Figure 1. State of Florida and the approximate study area surrounding Tampa Bay.

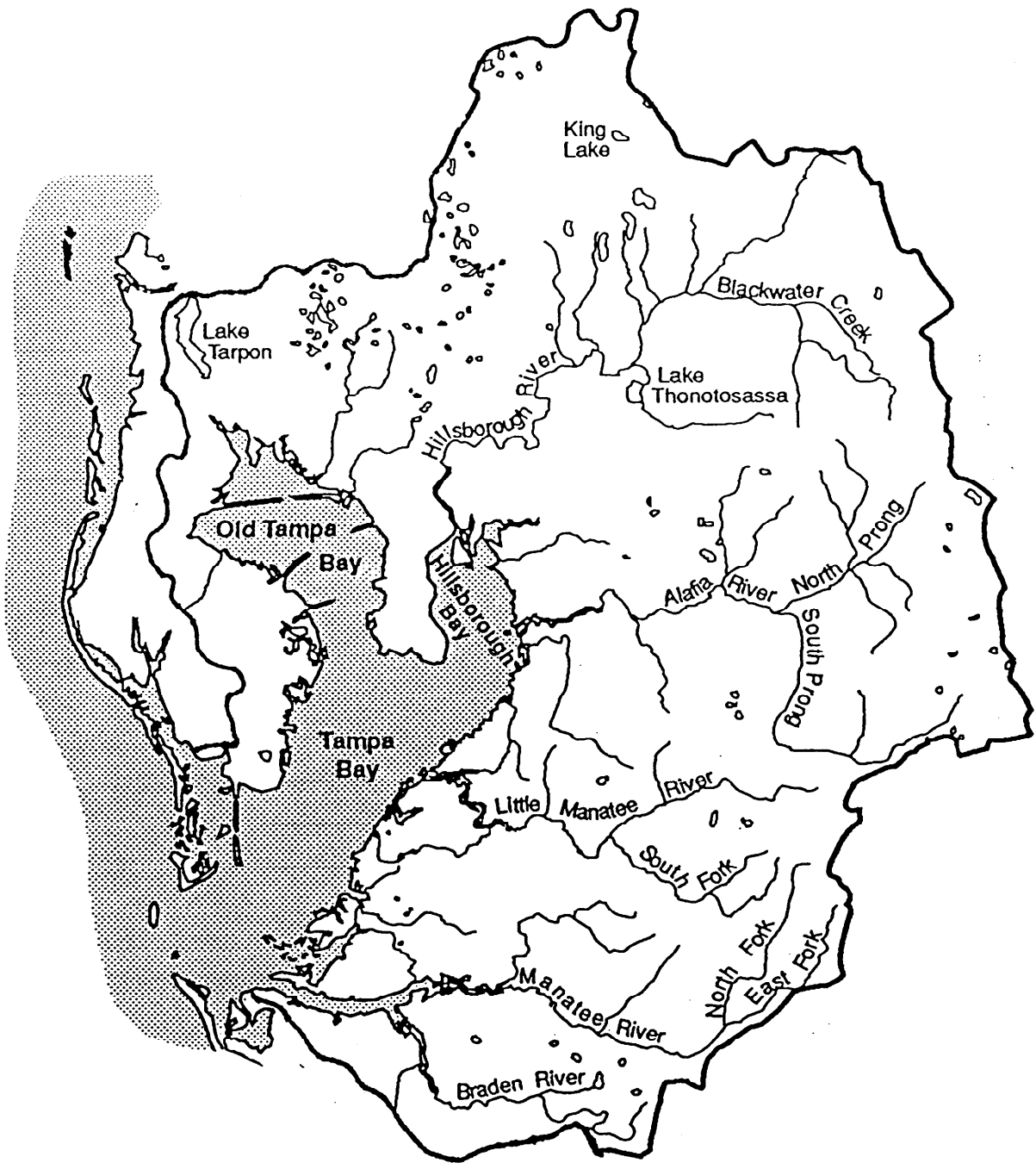


Figure 2. The watershed of Tampa Bay, including portions of Boca Ciega Bay (adapted from Wolfe and Drew, 1990).

## Study Area Emission Sources

Permitted stationary sources are listed in county and state emissions inventories. While  $\text{NO}_x$  may or may not be listed on permit conditions, estimates of annual  $\text{NO}_x$  emissions are prepared from standard factors based on fuel type, usage, or industry type. Advanced technologies for emissions reductions may not be reflected in the standard factors and so annual estimated emissions should be considered to be a conservatively high estimate of the total (Mr. Bob Soich, FDEP Tampa, personal communication).

A Florida emissions inventory for 1982 - 1983, including mobile sources, totaled 667,400 metric tons/year of  $\text{NO}_x$ . Statewide, as of 1982 (Pollman and Canfield, 1993), annual  $\text{NO}_x$  emissions (Table 1) were divided into:

Table 1. Allocation of total  $\text{NO}_x$  emissions in Florida by source (Pollman and Canfield, 1993).

Source Type	Percentage (%)
Point Sources	utilities 32
	non-utilities 21
Area Sources	transportation 43
	other 4

More recent compilations (Rogers, 1990) placed mobile sources responsible for nearer 50% of the total  $\text{NO}_x$  emissions, and indicated that Florida ranked eighth in the nation for  $\text{NO}_x$  emissions. Within the state in 1989, utilities accounted for 35 % of the total  $\text{NO}_x$  emissions with industrial sources contributing 5 - 10 %. Of the top 20 utility sources of  $\text{NO}_x$ , within the state during 1989, four are located within the Tampa Bay watershed, with an additional three in surrounding counties (Rogers, 1990).

As of 1992, FDEP operating report summaries (FDEP, 1994) indicated that stationary source  $\text{NO}_x$  emissions totaled 107,600 metric tons/year for the nine county area immediately surrounding the Tampa Bay and its watershed (Appendix A). The inclusion of Citrus County adds an additional 43,923 metric tons/year from the Crystal River Power Plant. Industry types represented on this listing include utilities and smaller power generation facilities, waste incineration, mining, and fertilizer processing and transportation. State listings by county are more complete for counties designated as non-attainment areas (Hillsborough and Pinellas), since counties without air quality violations have to report major emitters (> 90.8 metric tons/year [100 tons/year] of pollutants) only (Mr. Tom Rogers, FDEP, personal communication).

There are 15 stationary sources of  $\text{NO}_x$  greater than 454 metric tons/year (500 tons/year) within the nine county area surrounding the Tampa Bay watershed. Collectively, the emissions

represent 92% of the total stationary NO<sub>x</sub> emissions, and include a number of electric power generating stations and mining interests, as well as several resource recovery (incineration) installations. Figure 3 illustrates the locations of the sources with greater than 454 metric tons/year of NO<sub>x</sub> emissions and include Teco Big Bend, Teco Cannon, Florida Power Anclote, Florida Power & Light Manatee, City of Lakeland McIntosh Power, Florida Power Bar-tow, and Pinellas County Resource Recovery Facility. Major sources outside the watershed, but potentially affecting deposition quality to Tampa Bay also include Florida Power Crystal River and Florida Power and Lime (Brooksville).

Mobile sources of NO, are also substantial in urbanized, high traffic areas, estimated in Figure 4 by roadway density. Under the Clean Air Act Amendments of 1990, substantial reductions in emissions are required from the transportation sector and transportation plans and programs in areas where National Air Quality Standards are not achieved (non-attainment areas) must demonstrate conformity with State Implementation Plans (SIP), and contribute to annual emissions reduction. Emissions modeling and impact analyses have been performed by the Metropolitan Planning Organizations, to project the emissions reductions necessary to achieve compliance.

Ambient air modeling (with the Florida Standard Urban Transportation Modeling Structure and MOBILE 5.0) consists of an analysis of vehicle activity, average speed, miles traveled, and differing emissions characteristics for individual vehicle classes to compute estimates for the emissions of VOC (volatile organic compounds), CO (carbon monoxide), and NO<sub>x</sub> (nitrogen oxides). While the model has the ability to work in fairly small geographic segments, current analyses treat the entire county as a single entity. As a result, the modeling results are useful for assessing the relative magnitude of mobile sources with respect to stationary sources, but will not materially assist the Atmospheric Deposition Project in the quantitative identification of areas of poor air quality.

Counties lacking no non-attainment areas do not have the same level of modeling requirements and so projections are may not be available for the 1990 base year. Nevertheless, the table below (Table 2) summarizes some available information on transportation emissions of NO<sub>x</sub>. Data are in units of metric tons per day during the peak ozone season, and so multiplying the mobile emissions by 365 will give an approximate annual total. The annualized values total approximately 50,030 metric tons/year for the mobile emissions in the three county area immediately surrounding the Bay. Mobile sources, even with incomplete values, comprise approximately 30% of the total NO<sub>x</sub> emissions (stationary and mobile) within the nine county area surrounding Tampa Bay. Total NO, emissions in the study area account for nearly 24% of the State's estimated 667,400 metric tons/year of NO<sub>x</sub> emissions.

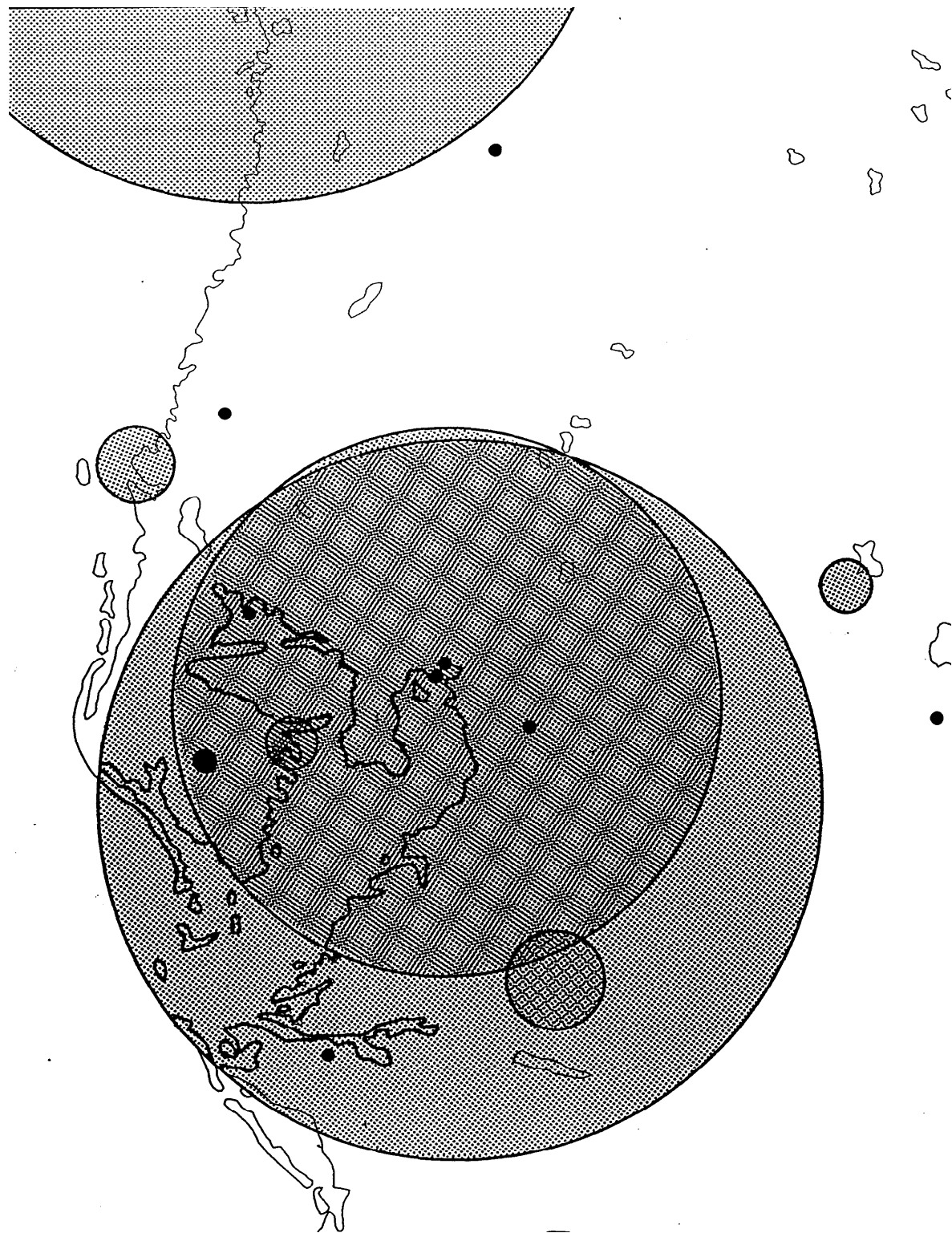


Figure 3. Stationary sources of NO<sub>x</sub> greater than 454 metric tons/year (500 tons/year). Center of the circle is at the location of the point source with the size of the circle proportional to the annual emissions.

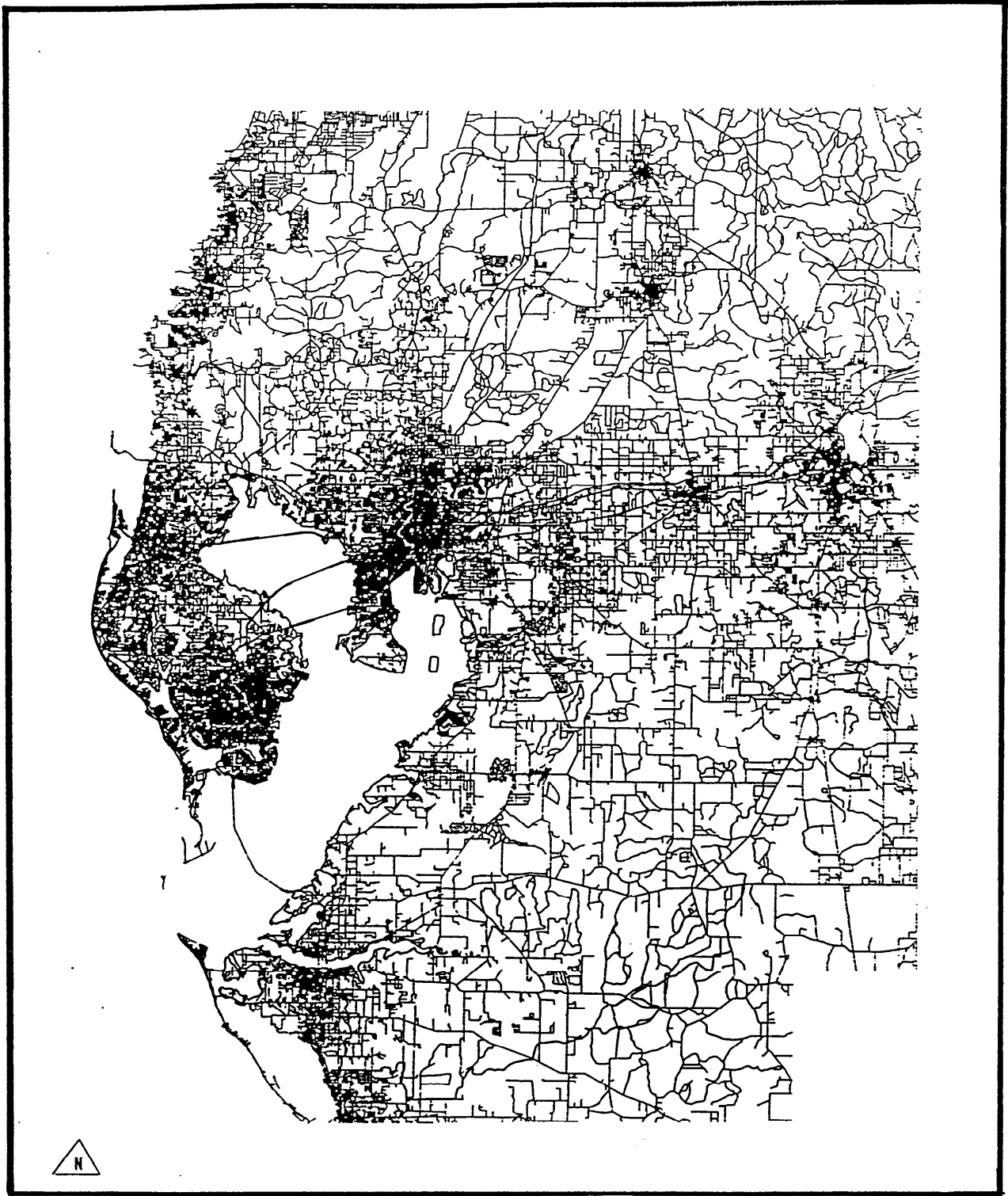


Figure 4. Roadway density within the Tampa Bay watershed, as a surrogate for areas of mobile emissions (Dames & Moore, 1990).

Table 2. Approximate emissions from mobile sources in counties adjacent to Tampa Bay.

County	VOC	CO	NO <sub>x</sub>	Date
	mton/day	mton/dy	mton/dy	
Hillsborough County	81.5	630	63.9	1990 <sup>a</sup>
	99.6	728	78.8	1990 <sup>b</sup>
Pinellas County	69.3	538	46.4	1990 <sup>a</sup>
Manatee County	14.5	109	11.8	1995-2010 <sup>a</sup>
Sarasota County	21.8	160	18.2	1995-2010 <sup>a</sup>

<sup>a</sup> Mr. Richard McElveen, FDEP, personal communication.

<sup>b</sup> HCMPO, 1993

Phosphorus has neither primary nor secondary air quality standards, but does appear on the listing of hazardous air pollutants. No facility within the nine county area, however, has any estimated or actual 1992 emission of this element (FDEP, 1994). With no stationary point sources, therefore, the emissions of phosphorus are primarily diffuse and non-point in nature.

Product shrinkage at bulk loading facilities of fertilizer products has been estimated between 0.01% and 1% of total shipments (Cardinale and Dunn, 1991; Johansson and Lewis, 1992) with the best estimates near 0.05%. Most recent computations from 1991 shipping records estimate that 13,225 metric tons/year of nitrogen and 47,826 metric tons/year of phosphorus are lost in shipping (Asci and Dames & Moore, 1993). Zarbock, *et al.*, (1993) estimated total fugitive emissions at near 1,700 metric tons/year of phosphorus for Hillsborough Bay and 272 metric tons/year for Lower Tampa Bay (Port Manatee) using a 0.05% rate for product shrinkage. Nitrogen loads for Hillsborough Bay and Port Manatee were estimated at 454,000 kg/year and 72,600 kg/year, respectively. Revisions to the fugitive emissions estimates are expected early in 1994.

The Air Program at Region IV of the U.S. Environmental Protection Agency (USEPA) is also interested in the phenomena. The Tampa Bay Initiative is examining the potential impact of the atmospheric deposition of both phosphorus and nitrogen. Initial reports summarizing existing information on phosphorus emissions are available, with information on nitrogen to be summarized during 1994 (Mr. Danny Orlando, USEPA, personal communication).

Much of the fugitive loss of phosphatic materials, however, is in surface runoff, or from the loss of large particle sizes which fall directly into Hillsborough Bay, rather than from the entrainment of small particulates into the atmosphere. As a particulate, even the smallest sized fraction is likely to be re-deposited relatively near the source, and while the ultimate impact on the waterbody is roughly equivalent, an unknown fraction of the material is small enough to be transported any substantial distance.



The remaining non-point sources of phosphorus emissions are the application of fertilizers or in soil resuspension from fields under cultivation. The size of these two sources is unknown, but since some 40% of the watershed is classified as agricultural, the magnitude could be substantial. Seasonal signals in crustal elements which coincide with spring tillage patterns have been observed elsewhere (Scudlark, *et al.*, in press).

Also in the FDEP 1992 annual operating reports (FDEP, 1994) are the estimated emissions for lead 46.2 metric tons/year) and mercury 1.59 metric tons/year (1.75 tons/year) for the sources listed within the nine county area surrounding the Tampa Bay watershed. Other estimates for mercury are substantially higher, KBN (1992) estimating 4.54 metric tons/year of mercury emissions primarily from solid waste incineration, utilities, and medical waste incineration. Additional and more specific information on sources of toxics within the Tampa Bay region will be available through the Toxics Contamination Action Plans.

In reviewing the emissions of various utilities and industries in the Bay area, it is apparent that several represent substantial sources. In addition to the size of the facility (megawatts produced), a controlling factor is the type of fuel used. Coal fired boilers are known to produce higher levels of NO<sub>x</sub> emissions, but in addition, coal contains higher levels of most trace metals than does fuel oil. Metals in solid wastes are even more enriched. Concentrations of selected metals in representative fuels appear below (Table 3) as tabulated by Windham (In press).

Table 3. Approximate trace metal content of different fuels (Windham, in press).

Trace Metal	#6 Oil	Eastern Coal	RDF/MSW
	µg/g	µg/g	µg/g
Copper	0.20	16	300
Lead	0.6	14	380
Zinc	0.8	40	600
Mercury	0.03	0.26	2

RDF/MSW - Refuse derived fuel/municipal solid waste

### Ambient Air Monitoring and Quality

Of the parameters of interest in this project, National Ambient Air Quality Standards (NAAQS) exist only for nitrogen dioxide (NO<sub>2</sub>) and for lead (Pb) (40 CFR Ch. 1. Subchapter C, Part 50). The annual ambient air standard for NO<sub>x</sub> is 100 µg/m<sup>3</sup>, or 53 ppb as an annual arithmetic mean. The other forms of nitrogen oxides or nitric acid are not directly regulated, although NO<sub>x</sub> emissions are inventoried yearly (FDEP, 1993) and compared against permitted levels. Lead standards are set at 1.5 µg/m<sup>3</sup>, calculated as the maximum arithmetic mean averaged over a calendar quarter.

The Tampa Bay airshed is designated as an area of non-attainment for ozone, in that 120 ppb or 235  $\mu\text{g}/\text{m}^3$  of ozone (as an hourly average) occurs during more than one day. There is also an area that is non-classifiable for lead in Hillsborough county, although 1992 data indicate the region may now be in compliance, with maximum quarterly averages of 1.4  $\mu\text{g}/\text{m}^3$  (FDEP, 1993) and an instantaneous maximum of 2.3  $\mu\text{g}/\text{m}^3$  (HCEPC, 1993). These areas of non-attainment are responsible for the relative density of monitors for ozone, ozone precursors, and lead within the Bay region.

Monitors of  $\text{NO}_x$  within Florida do not generally have a sufficient data base to determine long-term trends in ambient  $\text{NO}_x$  concentrations, but there are no short term trends apparent in the four monitors in Palm Beach, Dade, and Duval Counties for the period 1988- 1992 (FDEP, 1993b). Data from Hillsborough County indicate that annual  $\text{NO}_x$  averages have improved slightly since sampling began in 1975 and that no violations of NAAQS occurred during 1992 (HCEPC, 1993).

Pinellas County operates two ambient air sites with  $\text{NO}_x$  monitoring equipment (Figure 5). The Azalea Park site in the Tyrone area has a period of record from 1978-1985 and from 1988-present. As of 1991, wind speed and direction are also available at this location. (Mr. Tom Stringfellow, personal communication). A second site near east Lake Tarpon, at the John A. Chesnut Sr. Park, has also been monitored for  $\text{NO}_x$  since 1988, with wind data available since 1970. There are no current plans to add additional  $\text{NO}_x$  monitors. Lead monitoring is conducted in three locations, at east Lake Tarpon, at Tyrone Square, as well as at the Sheriff's Office on Ulmerton Road. Wind data are also available from the Tarpon Springs site (Anclote and Brady Roads) and the Derby Lane site at the western end of the Gandy Bridge.

Nitrogen dioxide data indicate that in 1992 the Azalea Park site had both higher maxima (72 versus 50 ppb) and higher average ambient concentrations (12 versus 7 ppb) than the east Lake Tarpon site (PCDEM, 1993a). Composite monthly means of lead data indicate that no substantive violations of lead criteria have existed since 1986 (PCDEM, 1993b) and all quarterly averages during 1992 were reported as 0.0 (FDEP, 1993). Wind direction for 1992 in Pinellas County is illustrated in Figure 6 with winds from the northeast to east dominating, followed by winds from the northwest and north/northwest. Collectively, winds from these directions represent about 40 % of the data points. Additional wind data from Tampa International Airport (PCDEM, 1993a) and other locations around the Bay, however, illustrate the regional nature of prevailing wind direction, and the difficulty of extrapolating wind (and therefore emission plume) directions to other regions of the Bay area.

Hillsborough County has 11 continuous air monitoring sites, but only one, on Gandy Boulevard by Tampa Bay, measures  $\text{NO}_x$  (Figure 5). The county is in attainment for this pollutant (HCEPC, 1993), with no exceedances of the standard in 1992, and an annual arithmetic mean of 10.0 ppb. Concentrations of  $\text{NO}_2$  have also improved slightly with time since 1975, indicating an improvement in mobile source control. Wind speed and direction are also recorded at the Gandy site, which has been in operation since April 1991. (Wind data are recorded at Davis Island and Simmons Park, as well.) An  $\text{NO}_x$  data base is apparently available for Davis Island, but the site is no longer monitoring this parameter. A future  $\text{NO}_x$  installation is planned

for the Simmons Park monitoring site on the southeastern shore of Tampa Bay (Mr. George Frader, HCEPC, personal communication).

Lead monitors within Hillsborough County are relatively numerous, due to the county's marginal non-attainment status for this pollutant. Six sites are designated as Health Department, Ruskin, Seminole, Johnson Ctrl, Gulf Coast, and NDC. Several are designated as special purpose monitors and sited to measure resuspension near lead point sources. The highest quarterly average for lead parameter in Hillsborough County was  $1.4 \text{ ug/m}^3$  in 1992 at 1700 N. 66<sup>th</sup> St. Trends in sources of lead indicate that transportation now contributes only 25 % of the annual emission, down from 73 % in 1985 (HCEPC, 1993).

Manatee County operates a single station for  $\text{NO}_x$ . The installation has been in place since October 1992, but is part of a Citrus Bum Study and is temporary, with operation scheduled only through October 1994 or 1995 (Mr. Rob Baum, Manatee County Environmental Action Commission, personal communication). The site is located east of the Bay on Highway 62, near the FPL Manatee Plant in Parrish. Mean  $\text{NO}_x$  concentration was 9 ppb for 1992 (FDEP, 1993). Windspeed, direction, and rainfall are also recorded. An annual report on the first year of data collection is scheduled for production this year. The data from the four sites in the watershed varies slightly, with an annual average high of  $16 \text{ ug/m}^3$  in southern Pinellas County to a low of  $9 \text{ ug/m}^3$  in Manatee County. Insufficient stations exist to map gradients for most of the watershed.

Based on the 1992 FDEP annual operating reports (FDEP, 1993), no other  $\text{NO}_x$  monitors are operating in Pasco, Polk, or Sarasota Counties. There are no monitors in Citrus, Hernando, Sumpter, Hardee, Desoto, or Charlotte Counties. Data are available through FDEP, although a change in storage format may make the retrieval of older data more problematic (Mr. Brian Kerckhoff, FDEP, personal communication). In addition, meteorological data such as wind speed and direction are not typically forwarded to FDEP, necessitating retrieval of this information from the individual county air programs. In some cases, meteorological data are not retained as magnetic or digital files.

Until recently, electric utilities were not required to continuously monitor emissions. Operating reports were prepared from standard emission factors of emissions per Btu generated with annual compliance sampling conducted at full load. With the Clean Air Act Amendments, however, utilities will be adding continuous monitors for a number of parameters, including  $\text{NO}_x$ , which will provide emissions data in real time. Most will be in place by a 1995 deadline (Mr. Ken Hedrich, FPC, personal communication).

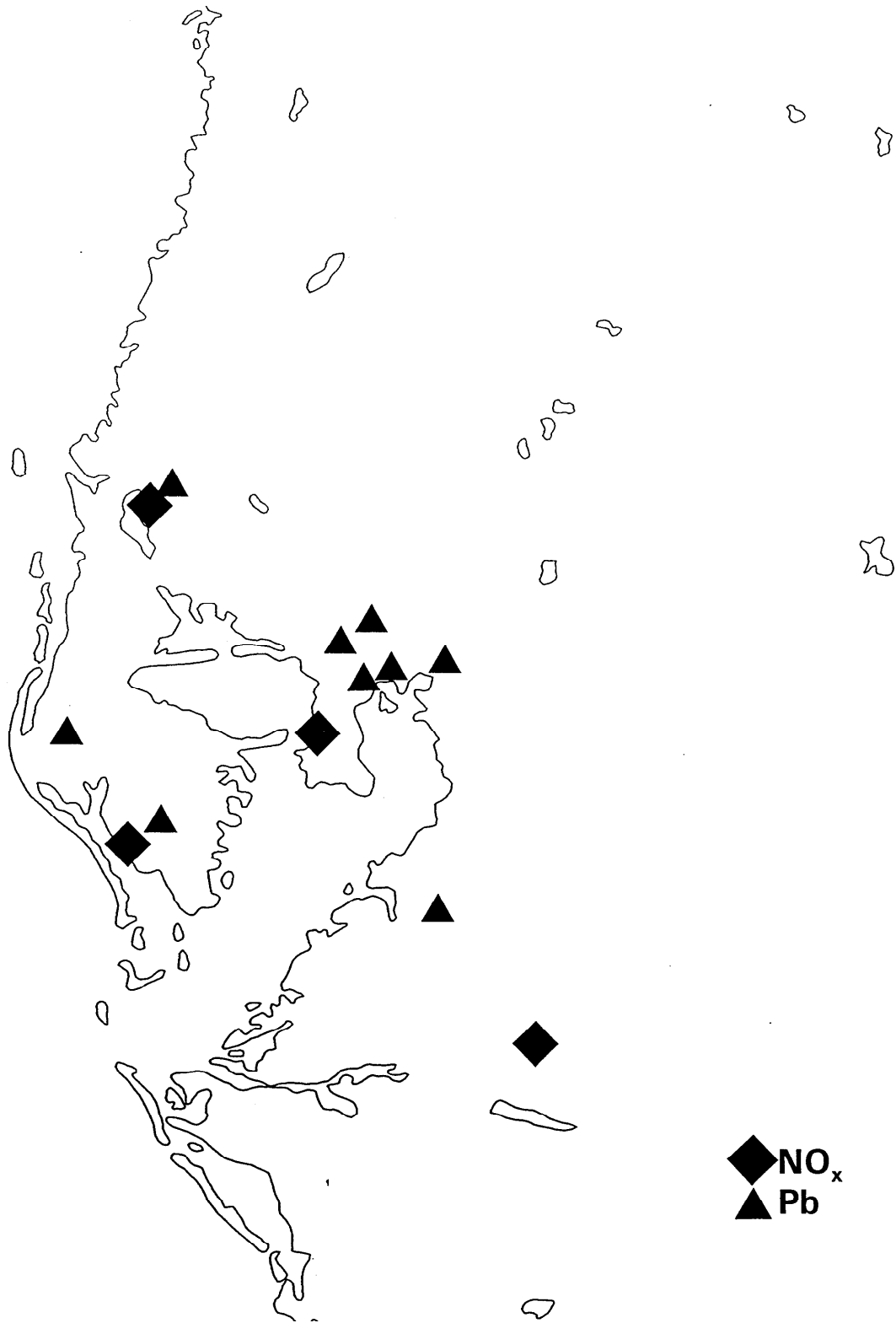


Figure 5. Locations of Hillsborough, Manatee, and Pinellas County NO<sub>x</sub> and lead ambient air monitoring stations.

wrplot91

January 1-December 31; Midnight-11 PM

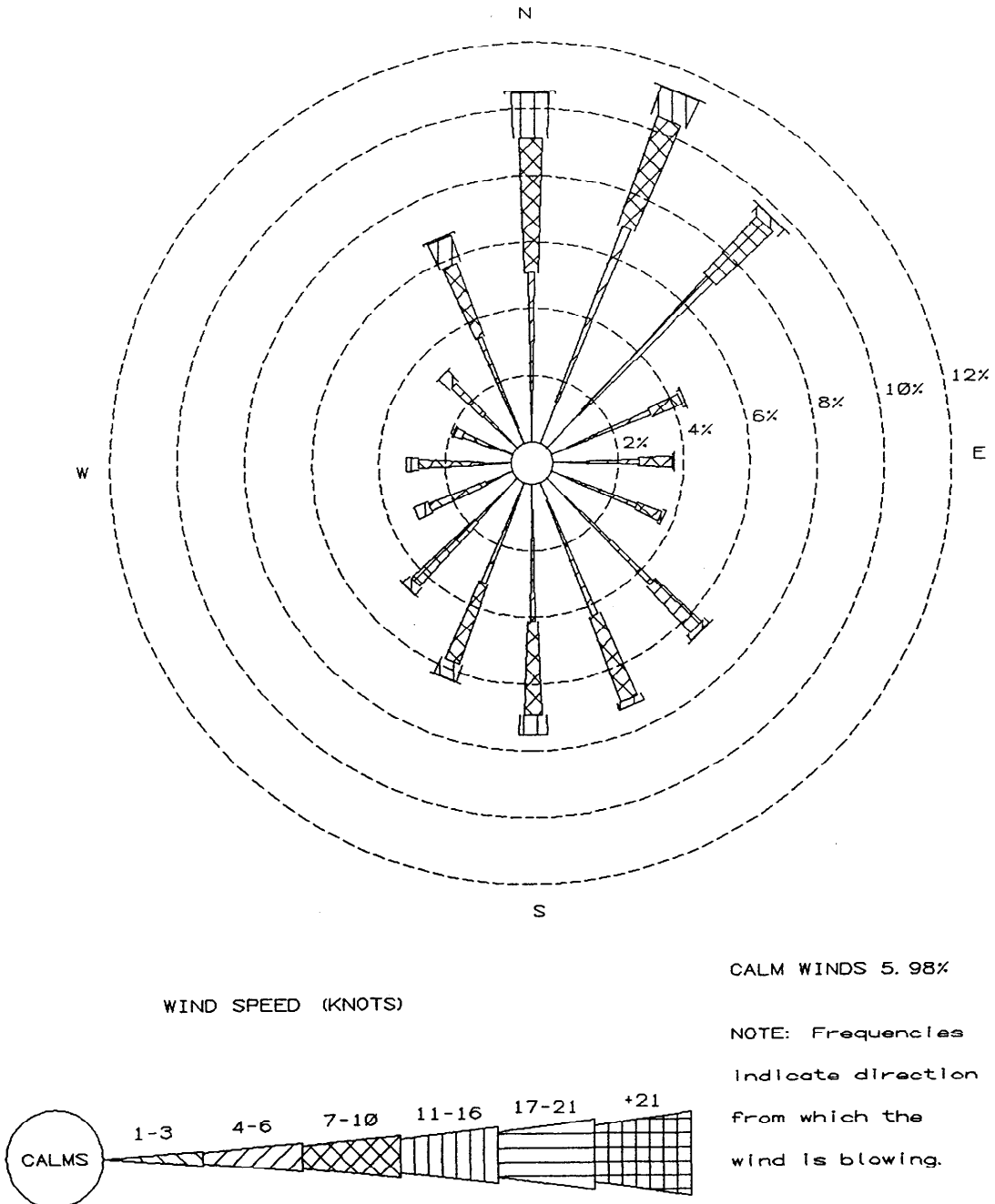


Figure 6. Wind direction frequency and strength, Pinellas County, 1991.

## Precipitation Chemistry

Precipitation chemistry has a history in Florida which extends at least to the mid 1950's (Junge and Werby, 1955), with periodic additional work. Data from Tampa were also gathered by Lodge, *et al.*, (1968) and Edgerton and Brezonik (1981). The primary focus of this discussion, however, will be on either more recent data or on networks or studies with a longer record.

In general, seasonal variations in rainfall quality are apparent for Florida, with maximum concentrations of nitrate ( $\text{NO}_3^-$ ), ammonium ( $\text{NH}_4^+$ ), and other major ions typically observed between May and October (Pollman, 1993; Madsen, *et al.*, 1990; NADP/NTN data for Verna Wellfield). Higher rainfall amounts during these months result in higher wet depositions during the summer months. Seasonality could be the result of a combination of higher power usage during warmer months (and therefore higher utility emissions), summer circulation patterns, or a proposed more effective scavenging mechanism for the convective storms which characterize the summer rainfall events (Pollman and Canfield, 1993). Geographically, while  $\text{SO}_4^{2-}$  and  $\text{H}^+$  concentrations in rainfall decrease from north to south within Florida,  $\text{NO}_3^-$  concentrations were more constant among the rural sites (Hunter/ESE, 1989), with wet deposition ranging from 1.68 to 2.25 kg/ha/year of nitrogen (Pollman and Canfield, 1993). Ammonium concentrations in precipitation apparently are controlled by local factors (Pollman and Canfield, 1993).

A number of programs have examined precipitation chemistry in Florida, with the longest term studies conducted by the National Atmospheric Deposition Program/National Trends Network (NADP/NTN) and the Florida Acid Deposition Study (FADS). The NADP/NTN network operates five sites in Florida, with the closest to Tampa Bay located at the Verna Wellfields in Sarasota County. Other sites are at the NASA facility near Cape Canaveral, in the Everglades, near the Georgia border, and near the Alabama border. Parameters include major anions and cations with orthophosphate available by special request. Nitrate-nitrogen wet-only deposition from the NADP sites through 1989, as summarized by Winchester and Fu (1992), appears below (Table 4), and nitrate concentrations can be seen to converge near 0.22 mg/l.

Table 4. Mean precipitation concentrations and deposition of nitrogen for the Florida NADP/NTN sites (Winchester and Fu, 1992).

Site	Mean NO <sub>3</sub> -N Concentrations	Wet Deposition
	mg/l	kg N/ha/year
Dade	0.145 +/-0.008	0.96+/-0.08
Verna	0.223 +/-0.016	1.45+/-0.18
Kennedy	0.210+/-0.013	1.43+/-0.13
Bradford Forest	0.226+/-0.010	1.68+/-0.09
Quincy	0.230+/-0.018	1.53+/-0.15

The FADS work, and subsequent Florida Acid Deposition Monitoring Program (FADMP), consisted of up to 14 stations which operated between 1981 and present. Siting criteria placed stations in selected rural areas to evaluate regional and temporal trends, develop source attributions, and evaluate ecological effects of acid precipitation in Florida. At times, wet samples were collected on both daily and weekly schedules, varying by station and the year of the study. Parameters included major ions (with  $\text{NO}_3^-$  and  $\text{NH}_4^+$  and physical parameters. Dry samples were collected every two months, but only analyzed between 1981 and 1982. Ambient air concentrations were also collected, including analyses for  $\text{HNO}_3$  and  $\text{NO}_2$ . Currently, the network consists of two sites, one near Careyville in the panhandle area and the other at Archbold Biological Research Station in Highlands County. Support for the project was provided by the Florida Electric Power Coordinating Group (FCG) with the Florida Department of Environmental Protection scheduled to assume operation of the remaining two sites in the near future.

The closest FADS station to the Tampa Bay watershed was in Zephyrhills, in Pasco County, with weekly data collected at the site from 1981 through 1990. Daily data were gathered at the site during the 1982-83 sampling year. Ambient air data were collected at least between 1982 and 1984. Data on selected metals (vanadium, aluminum) and phosphate were collected during the initial sampling year. The raw data are not currently available, but extracts from selected annual summaries are provided below.

Statewide, while acidic deposition and  $\text{SO}_4^{-2}$  decreased from north to south, no significant pattern was seen in  $\text{NO}_3^-$  or  $\text{NH}_4^+$  (ESE, 1986). Volume weighted mean concentrations for the Zephyrhills site between 1981 and 1984 and seasonal differences during 1988 (Hunter/ESE, 1989) are presented below (Table 5). Interannual variations in weighted  $\text{NO}_3^-$  concentrations were near 10% during 1981-1984 (ESE, 1986). Seasonal differences in rainfall also indicate that the majority of inorganic nitrogen would be deposited during the summer (Hunter/ESE, 1989). The central Florida stations, including the Zephyrhills site, had the highest nitrate weighted means. Phosphate volume weighted means averaged  $< 0.01$  mg/l  $\text{PO}_4\text{-P}$  statewide, with the maximum value (0.04 mg/l  $\text{PO}_4\text{-P}$ ) at the Zephyrhills station. This result was thought to reflect nearby mining operations.

Table 5. Volume weighted mean rainfall concentrations and wet deposition values for the FADS Zephyrhills site (ESE, 1986).

Chemical Species	-----Volume Weighted Mean Concentrations-----		
	1981-1984	1988 Summer	1988 Winter
	mg/l	mg/l	mg/l
NH <sub>4</sub> -N	0.133	0.158	0.124
NO <sub>3</sub> -N	0.166	0.207	0.088
PO <sub>4</sub> -P	0.043*		

\* - Data from 1981-1982

	----- Wet Deposition -----	
	1981-1984	1988
	kg/ha/year	kg/ha/year
NH <sub>4</sub> -N	1.67	1.89
NO <sub>3</sub> -N	2.07	2.06
PO <sub>4</sub> -P	0.50	

It was determined that data from the dry samples collected under FADS could not be used for estimates of dry deposition, and so analyses were terminated on this sample type after the initial year. To determine dry deposition rates, ambient air concentrations were determined for approximately one year at the Zephyrhills site between 1982-1984. Concentrations averaged 0.244 ug N/m<sup>3</sup> for HNO<sub>3</sub>-N and 3.26 ug N/m<sup>3</sup> of NO<sub>2</sub>-N (1.1 and 10.7 ug/m<sup>3</sup> of HNO<sub>3</sub> and NO<sub>2</sub>, respectively). Data from other sites in 1988 suggest that the bulk of the ambient nitrogen was in the form of NO<sub>2</sub>, followed by HNO<sub>3</sub>, with minimal quantities as aerosol NH<sub>4</sub><sup>+</sup> and NH<sub>3</sub> (Hunter/ESE, 1989). Total wet and dry deposition of inorganic nitrogen at the Zephyrhills site, excluding NO<sub>3</sub><sup>-</sup>, NH<sub>4</sub><sup>+</sup>, and NH<sub>3</sub>, was estimated as 6.13 kg/ha/year.

Brezonik, *et al.*, (1981) also presented loading rates for Florida. As summarized in Blancher and Stewart (1991), estimates for urban and rural bulk loading of nitrogen were 7.6 and 5.8 kg/ha/year, respectively. Total bulk deposition direct to the Bay was estimated by Blancher and Stewart (1991), using the coastal bulk precipitation coefficients of Brezonik, *et al.*, (1981) at over 521 metric tons/year of nitrogen and near 28 metric tons/year of phosphorus.

Madsen, *et al.*, (1990) have also conducted a lengthy study of acid rain at two locations, Orlando (University of Central Florida) and the Kennedy (Canaveral) Space Center. The period of record is at least 1981- 1990, with only minor interruptions in data collection, during which the Kennedy site was moved slightly and incorporated into the NADP/NTN network. Extreme variations in daily and monthly rainfall quality were noted, with moderate annual variations.



Integrated wet samples of 24 hour, 72 hour, and one week periods were collected with wet deposition values (Table 6) averaging:

Table 6. Central Florida wet deposition values (Madsen, *et al.*, 1990).

Chemical Species	UCF 1977-89	KSC 1977-81	NADP 1984-89
	kg/ha/year	kg/ha/year	kg/ha/year
NH <sub>4</sub> -N	1.76	1.85	0.98
NO <sub>3</sub> -N	2.94	1.93	2.04

UCF - University of Central Florida  
 KSC - Kennedy Space Center  
 NADP - NADP site at Kennedy Space Center

Allen and Sutton (1990) reported on a year of event-based wet deposition and triple filter pack samples for dry deposition collected at three sites in the Cary Forest, northeast of Gainesville, FL. Major inorganic ions were analyzed, including phosphorus. Total deposition was 6.6 kg N/ha/year. Nitrate concentrations in volume weighted precipitation were 0.21 mg N/l with NH<sub>4</sub><sup>+</sup> concentrations at 0.15 mg N/l. Phosphorus loadings (from phosphate) in rainfall were much lower, 0.013 mg P/l or 0.15 kg P/ha/year. Earlier work by Hendry and Brezonik (1980) and nearby NADP data as presented in Allen and Sutton (1990) also found similar precipitation concentrations (Table 7) for a central Florida region near Gainesville.

Table 7. Central Florida rainfall concentrations, as summarized by Allen and Sutton (1990).

	Hendry and Brezonik (1980) 1976-77	NADP Bradford Forest 1987	Cary Forest Allen and Sutton (1990) 1988-89
	mg/l	mg/l	mg/l
NH <sub>4</sub> -N	0.10	0.08	0.15
NO <sub>3</sub> -N	0.19	0.15	0.21
HPO <sub>4</sub> -P	0.02	- -	0.01

For the Carey Forest, rainfall was the dominant mode of deposition for nitrate and ammonium ions, with 37 % of the total nitrogen deposition attributable to dry deposition. Nitric acid vapor generally exceeded concentrations of particulate nitrate and nitrogen dioxide by a ratio of 3:2:1, respectively. Average annual wet and dry loadings are shown below (Table 8). Average annual dry deposition fluxes were calculated from theoretical deposition velocities and the ambient air concentrations and total (wet plus dry) deposition (less dry deposition of  $\text{NH}_3$  and  $\text{NH}_4^+$ ) was a near 6.62 kg/ha/year.

Table 8. Wet and dry deposition rates, Carey Forest, Florida (Allen and Sutton, 1990).

Chemical Species	Wet deposition
	kg/ha/year
$\text{NH}_4\text{-N}$	1.71 +/- 0.02
$\text{NO}_3\text{-N}$	2.48 +/- 0.05
$\text{HPO}_4\text{-P}$	0.15 +/- 0.04
	Dry Deposition
	kg/ha/year
particulate $\text{NO}_3\text{-N}$	0.12 +/- 0.01
$\text{HNO}_3\text{-N}$	2.20 +/- 0.24
$\text{NO}_2\text{-N}$	0.11 +/- 0.03

#### Precipitation and Loading Data for Tampa Bay

During the early 1980's, the Nationwide Urban Runoff Program (NURP) collected rainfall quality information at four sites within metropolitan Tampa (Noel, *et al.*, 1987) (Table 9). Although spanning an eight month period, the combined data only represent some 35 cm (approximately 14 inches) of rainfall per site and storms generally were greater than 0.5 cm before data analyses were complete. On an event basis, small storms generally represent a continuum of concentration values for all parameters, but larger storms have lower concentrations. Annual loads would likely be underestimated from this data set. Arithmetic and volume weighted means of Tampa NURP rainfall concentrations are listed below. The large standard deviations emphasize the episodic nature of atmospheric loadings. The data set is also valuable in that organic nitrogen and total phosphorus were analyzed, unlike most deposition networks. It is apparent that nearly half of the nitrogen in the NURP precipitation data set could be in the organic form and that phosphorus is almost all orthophosphate.

Table 9. Arithmetic and volume weighted mean rainfall concentrations collected under the NURP program in Tampa, FL (Noel, *et al.*, 1987).

Chemical Species	Mean	Std. Dev.	Volume Weighted Mean
	mg/l	mg/l	mg/l
Organic N	0.45	0.49	0.42
NH <sub>3</sub> -N	0.12	0.08	0.12
NO <sub>2</sub> -N	0.02	0.05	0.01
NO <sub>3</sub> -N	0.45	0.74	0.33
TKN	0.50	0.52	0.47
Total P	0.17	0.16	0.16
PO <sub>4</sub> -P	0.12	0.14	0.09
Cadmium	0.001	0.001	0.001
Copper	0.043	0.031	0.036
Lead	0.009	0.011	0.006
Zinc	0.154	0.094	0.124

A subset of the NURP precipitation data was subsequently used by Hartigan and Hanson-Walton (1984) in estimating loadings to Tampa Bay. Interestingly, the mean rainfall concentrations of total nitrogen used ranged between 53 % and 103 % of the mean concentrations used as loading factors for stormwater runoff from the various land uses, indicating that a substantial fraction of runoff nitrogen originated as wet deposition. Phosphorus showed similar patterns. Wet deposition accounted for 43 % - 113 % of the total phosphorus in runoff.

The Pinellas County Air Quality Division established a wet deposition monitoring site at Cross Bayou, near the St. Petersburg/Clear-water Airport, and approximately 60 meters from U.S. Highway 19. This site was operated during 1984-1989 and 1991-1992 under NADP/NTN protocols, with analyses conducted for major ions (including nitrate and ammonia) and physical parameters (PCDEM, 1993c). Approximately 360 valid samples were collected and annual means and loadings presented below (Table 10). A review of the annual averages emphasizes the difficulty of extrapolating loading rates from short-term data sets.

Table 10. Volume weighted mean rainfall concentrations and wet deposition at Cross Bayou in Tampa, FL (PCDEM, 1993c).

Year	----- Volume -----		----- Wet -----	
	Weighted Mean NO <sub>3</sub> -N	NH <sub>4</sub> -N	Deposition NO <sub>3</sub> -N	NH <sub>4</sub> -N
	mg/l	mg/l	kg/ha/year	kg/ha/year
1984	0.338	0.474	1.19	1.68
1985	0.057	0.226	0.19	0.74
1986	0.089	0.391	0.27	1.18
1987	0.204	0.165	3.56	2.89
1989	0.156	0.123	1.59	1.25
1991	0.286	0.127	2.56	1.14
1992	0.271	0.097	2.64	0.95
Mean	0.200	0.229	1.71	1.40

Analyses of the 1985 and 1986 Cross Bayou data sets (Molesch, 1991) identified higher rainfall acidity (lower Ph) during the summer months when weekly rainfall amounts are lower and emissions of acid rain precursors (sulfur dioxide and nitrogen dioxide) are higher. Nitrate and sulfate depositions were directly proportional with 84% of the total estimated to be from anthropogenic sources. Summer convective storms were observed to account for seasonal increases in nitrate deposition. Acidic components of deposition in this study were quite comparable to values observed for Florida and Georgia (PCDEM, 1993c). The annual wet deposition loadings for nitrate-nitrogen are comparable to those determined by Allen and Sutton (1990), Brezonik, *et al.* (1980), Madsen, *et al.*, (1990) and NADP data from Winchester and Fu (1992). Annual loads for NO<sub>3</sub><sup>-</sup> and NH<sub>4</sub><sup>+</sup> combined totaled 3.11 kg/ha/year.

A few bulk deposition samples were also collected as part of a 1990-1991 study of stormwater discharge to Lake Tarpon (CCI, 1992). Samples from the one station were analyzed for nutrients (nitrogen and phosphorus), solids, and biochemical oxygen demand. Variability among the individual samples was high. Once an outlier sample was discarded for contamination, 50-60% of the total phosphorus was received as orthophosphate, organic nitrogen dominated the deposition of nitrogenous materials, more than two times the rate of NO<sub>x</sub> species, which in turn was nearly five times the rate of ammonia deposition. The NURP data (although wet only samples) exhibited a smaller fraction of total phosphorus as orthophosphate and a smaller proportion of organic nitrogen in comparison to nitrate-nitrogen. Bulk loading rates in the CCI (1992) study for total nitrogen in wet deposition were on the order of 10 kg N/ha/year.

More recently, and still ongoing, the Southwest Florida Water Management District (SWFWMD) is performing quality analyses on rainfall (wet deposition only) as part of a stormwater research program (Table 11). To date, data are available for a total of 94 events at

up to three sites, over a three year period, 1990-1993 (Rushton, 1993). Sites are located at Hidden River Corporate Park near Fletcher Avenue and I-75, at the Tampa office of SWFWMD on U.S. 301, and at Al Lopez Park (formerly Horizon Park) north of the Tampa Stadium between Dale Mabry and Himes Avenues. In general, only storms with greater than 0.75 cm of precipitation are retained for analysis, which would tend to underestimate annual loads. In this work, rainfall has been identified as a major source of inorganic nitrogen, exceeding stormwater runoff concentrations by two to four times. Spatial differences between three sites in Tampa have been detected for ammonia and zinc, while  $\text{NO}_{2+3}\text{-N}$  exhibits seasonal differences, being higher during the summer. The Tampa office site, nearest a major transportation corridor and feed lots, recorded the highest values of all parameters. Phosphorus data are drawn from Rushton (1991).

Table 11. Arithmetic and volume weighted mean rainfall concentrations collected under SWFWMD stormwater research programs (Rushton, 1991, 1993).

Chemical Species	Mean	Std. Dev.	Volume Weighted Mean
	mg/l	mg/l	mgll
$\text{NH}_3\text{-N}$	0.137	0.362	0.120
$\text{NO}_{2+3}\text{-N}$	0.222	0.490	0.193
Organic N	0.15	0.45	0.12
$\text{PO}_4\text{-P}$	0.031	0.233	0.022
Total P	0.055	0.332	0.042
Zinc	<0.03	0.16	< 0.01

Atmospheric nutrient loadings determined for Lake Maggiore in northern Pinellas County (CH<sub>2</sub>M Hill, 1991) employed a combination of NADP/NTN data, Pinellas County's data from Cross Bayou (DEM, Pinellas County, 1993), and Rushton's (1993) earlier data. As reported in King (1992) a factor was used to estimate dry deposition from wet data, with wet deposition of  $\text{NO}_x$  and  $\text{NH}_4^+$  assumed to equal 70% of the total (wet plus dry) deposition.

### Work in Progress

One of the most comprehensive studies on deposition of trace metals is currently underway, focussing on atmospheric mercury as well as trace metals, major ions and nutrients. The Florida Atmospheric Mercury Study (FAMS), conducted with support from the Florida Department of Environmental Protection (FDEP), the Electric Power Research Institute (EPRI), and Florida Power and Light (FPL), has five sites located in predominantly rural areas collecting both wet, bulk, and aerosol samples. The first installation was at Lake Barco near Gainesville in May 1992 and approximately a year of sampling has been conducted. The annual report (Landing, *et al.*, 1993) consists of explicit site installation and collection procedures and methodologies.

Subsequent sites which have been installed or are scheduled for deployment include stations near Fort Myers (downwind of a scheduled municipal incinerator), in the Fakahatchee Strand, and two in the Everglades, with two other sites potentially to be established in the Florida Keys and in Fort Lauderdale. While data reports from this program are not yet available, personal communication (Dr. William Landing, Florida State University) indicates that bulk deposition of metals ranges between one and five times that in wet deposition only.

In other work for SWFWMD, water and nutrient budgets are being prepared for Lake Panasofkee, just north of the Tampa Bay watershed. Monthly bulk deposition data collected as part of this project is not yet available (Dr. Walter Ogburn, CH<sub>2</sub>M Hill, personal communication), but the work has reportedly had occasional problems with contamination of bulk samplers from insects and bird droppings. In the South Florida Water Management District (SFWMD), additional deposition work is underway with 10-15 wetfall collectors and a total of 20 scheduled by the end of 1994 (Dr. James Grimshaw, SFWMD, personal communication). The network has been in place for some time, but earlier data consisted of bulk deposition values and, due to contamination and other problems, the older data are considered less useful than the wet deposition data collected more recently.

Pinellas County Water Resources Management Section, in conjunction with the United States Geological Survey (USGS), has recently begun a stormwater study at a domestic waste treatment facility near the northern end of the new Bayside Bridge over Old Tampa Bay. A component of the study includes weekly integrated samples of both dry, wet, and bulk deposition. While funding was sought to analyze trace metals and other toxics in the deposition samples, current analytical regimes do not include trace metals, but do include inorganic nitrogen series and total Kjeldahl nitrogen. The proportion of ammonia to nitrate nitrogen in the deposition samples will be of particular interest in this study, to assess the potential for ammonia releases from the treatment facility.

The USGS has also been involved in an extensive pilot study of the deposition of nitrogen and phosphorus to the surface of Lake Okeechobee through collection of wet/dry samples and bulk samples to wet and dry surfaces. These data, while very pertinent to the Tampa Bay study, are not yet available (Dr. Jake Peters, USGS, personal communication). The USGS is also currently collecting bulk precipitation samples in Hardee and Desoto Counties, in the Horse Creek watershed. The study is examining the use of chloride as a tracer for evaluating groundwater recharge. A limited suite of major ions does not include any nitrogen species, but the chloride data could be useful for evaluating the spatial variation in marine influences.

### Metals in other regions

While a number of researchers have collected data on the deposition of major ions in or near Tampa Bay, large data sets for trace metals in deposition are not numerous. Typically older data sets (pre-1982) are particularly suspect with regards to trace metal analytical accuracy due to subsequent improvements in quality assurance, handling, and analytical techniques. Metals data from Rushton (1993) are generally below detection limits. The data from the NURP program (Noel, et al., 1987), previously discussed for nutrients, also included metals analyses but concentrations are extremely high in relation to other reported values.

Due to the relative scarcity, data from regions outside Florida were also accumulated for first-order estimation purposes. Concentrations in precipitation may now exceed the aquatic concentrations in many ecosystems, and atmospheric inputs of trace metals exceed riverine inputs to the world oceans (Nriagu and Pacyna, 1988). Rainfall concentrations (wet deposition only) compiled by Nriagu (1991) appear below (Table 12).

Table 12. Compiled rainfall concentration of trace metals in urban and rural settings (Nriagu, 1991).

	-----Rainfall Concentrations-----	
	Urban	Rural
	$\mu\text{g/l}$	$\mu\text{g/l}$
Cadmium	0.35	0.05
Copper	2.8	0.45
Lead	6.0	1.4
Zinc	3.5	0.8
Mercury	0.065	0.025

Other authors (Windham, in press) present much higher levels of rainfall concentrations (perhaps with older data sets), but the pattern of urban concentrations higher than rural for all anthropogenic metals is consistent.

With sites in Colorado, Tennessee, and Illinois, Vermette, *et al.* (1992) presented wet deposition data for a 15 week period, and in addition summarized data from other programs (Great Lakes Atmospheric Deposition [GLAD] network) and other researchers (Nriagu, 1992) (Table 13). By comparison, the NURP data appear excessive, and it should be recalled that the data were collected in 1981 when analytical accuracy, together with potentially poorer air quality may have been a factor.

Table 13. A compilation of trace metal volume weighted means (Vermette *et al.*, 1992; Nriagu, 1992; Noel *et al.*, 1987).

	----- Volume / weighted / means -----				
	Cu	Pb	Zn	Cd	Hg
	$\mu\text{g/l}$	$\mu\text{g/l}$	$\mu\text{g/l}$	$\mu\text{g/l}$	$\mu\text{g/l}$
Vermette, <i>et al.</i> (1992)					
Colorado	1.32	1.49	4.05	0.07	0.0028
Illinois	1.33	1.11	3.06	0.10	0.0017
Tennessee	0.69	0.87	2.86	0.24	0.0013
GLAD	1.0	1.7	8.7	0.2	
Nriagu (1992) - rural	0.45	1.4	0.8	0.05	0.025
- urban	2.8	6.0	3.5	0.35	0.065
NURP	36	6	124	1	
GLAD: Great Lakes Atmospheric Deposition					

Orr, *et al.*, (1990) reported annual loadings for a number of metals in the Great Lakes region as monitored by the Acidic Precipitation in Ontario Study. Spatial trends were consistent with the presence of stationary point sources and ranged between 30-60 g/ha/year for lead, 20-60 g/ha/year for zinc, and 0.5-2.0 g/ha/year for cadmium. Loadings to the Great Lakes as determined by the GLAD network were approximately 20 g/ha/year for lead in 1987 (Klappenbach, E.W., 1990).

Church and Scudlark (1990) presented volume weighted mean concentrations and depositions for a number of metals in rainfall at Lewes, Delaware along the mid-Atlantic coast (Table 14). Mean values for the 1982-1989 time period are as follows, but the annual volume weighted mean for lead have declined to near 0.75 ug/l by 1989. Dry deposition of trace metals at Lewes were computed from an intensive series of aerosol measurements. Crustal components were normalized to aluminum, and deposition velocities used were those used for soil dusts (0.3 cm/set). Remaining elements were assumed to be associated with submicron aerosols, and a deposition velocity of 0.1 cm/set was used. Wet deposition, in general, dominated the total deposition of these elements.



Table 14. Trace metal volume weighted mean concentration and total deposition values for the mid-Atlantic region (Church and Scudlark, 1990).

Trace Metal	Vol. Weighted Mean	Wet Deposition	----- Dry Flux -----		Total Deposition
			Crustal	Non-crustal	
	µg/l	g/ha/yr	g/ha/yr	g/ha/yr	g/ha/yr
Cadmium	0.098	1.08		<0.01	1.1
Copper	0.760	8.36	0.19	1.3	9.86
Lead	1.9 (0.75 <sup>a</sup> )	20.9	0.01	19.7 <sup>b</sup>	40.7
Zinc	5.16	56.8	0.47	8.3	65.6

<sup>a</sup> 1989 Value

<sup>b</sup> Inaccurate value

Annual values for two sites on the shores of Chesapeake Bay were similar to the Delaware data in orders of magnitude (Scudlark, *et al.*, in press, Baker, *et al.*, 1992) (Table 15). Part of the Chesapeake Bay Atmospheric Deposition study (CBAD), the sites were primarily rural/agricultural but were downgradient from concentrated air traffic, coal fired power plants, and the Washington, DC/Baltimore, MD regions. Monthly depositions varied by a factor of 50 during the sampling year, with weekly fluctuations even larger. Intersite variation were largest for the weekly time scales. Most of the elements were determined, through aluminum data and crustal ratios, to be of anthropogenic origin. In comparison to fluvial inputs to Chesapeake Bay, direct atmospheric deposition contributes significant portions of lead, zinc, cadmium, and arsenic.

Table 15. Trace metal volume weighted mean concentration and total deposition values for the Chesapeake Bay region (Scudlark, *et al.*, in press; Baker *et al.*, 1992)

Trace Metal	Vol. Weighted Mean	Wet Deposition	Dry Deposition	Total Deposition
	$\mu\text{g/l}$	$\text{g/ha/yr}$	$\text{g/ha/yr}$	$\text{g/ha/yr}$
Cadmium	0.007	0.48	0.21	0.69
Copper	0.12	2.60	4.00	6.60
Lead	0.19	5.56	6.90	12.5
Zinc	0.39	13.35	20.0	33.00

### Dry Deposition

Dry deposition has been described above as a process which depends not only on the chemical reactivity of both atmospheric species and receiving surface, but also on the physical state (gaseous, particulate) and/or particle size of the material, and the micrometeorology and condition of the receiving surface (turbulence, roughness, surface moisture, type of vegetation, and temperature).

The currently accepted method to estimate total deposition is to collect wet-only samples with automated equipment and to determine dry deposition separately. Dry deposition, however, is not typically quantified with the dry side of the wet-dry automated sampler. A typical dry bucket sampler is expected to be a reasonable approximation of larger particulates which settle primarily by gravitation, but less so for aerosol sized or gaseous components (Baker, 1991). Accordingly, ambient air is sampled with a system of filters and pump to quantify the concentrations of selected species. The ambient air concentration of each species is subsequently multiplied by a literature deposition velocity to generate the assumed dry deposition for that species.

Deposition velocities, however, are not typically empirically determined, but are modeled parameters. The velocities are subject to many variables and uncertainties, and can vary by up to an order of magnitude, depending not only on the chemical species and its particle size, but also on season, temperature, time of day, surface wetness, chemical characteristics of the receiving surface, leaf area index, wind, and surface roughness, to name a few. Dry deposition has been directly measured in few laboratory and field settings and there is no accepted methodology for routine measurements of this parameter. Emphasis is placed on the highly variable nature (both temporally and spatially) of the process (Hunter/ESE, 1989). Spatial differences do not typically smooth out as longer periods of record are achieved (Hicks, *et al.*, 1992), as differences in the receiving surface remain fixed. There is no accepted methodology to routinely and directly measure dry deposition.

When deposition velocities are used for simulation, deposition rates are calculated from the deposition velocity,  $V_d$ , and the ambient concentration,  $c$  (dry deposition =  $V_d \times c$ ), lumping the effects of the wide variety of variables. Ambient concentrations are determined through a filter pack sampler, with or without a gradient of filter sizes. A range of deposition velocities have been determined, which can span from 0.04 to 0.72 cm/sec, a factor of 18, for equivalent sized particles (Baker, et al., 1992). Velocities determined for one surface are unlikely to be appropriate for others. While night-time velocities in terrestrial-based systems are reported to be negligible (Hunter/ESE, 1989), exchange during night (and winter) periods may dominate dry deposition to water bodies, whenever the water is warmer than the air (Hicks, et al., 1992). Wet surfaces are reported to be more efficient collectors of  $\text{NO}_3^-$  than a dry inert surface, while  $\text{NH}_4^+$  accumulates more on the dry surface (Lewis, 1983). For nitrogen, there is insufficient information to deduce velocities for some species such as  $\text{NH}_4^+$  and  $\text{NO}_2$ . In addition,  $\text{NH}_3$  may have a non-linear rate deposition rate which would be poorly simulated by  $V_d \times c$  (Edger-ton and Lavery, 1990).

It is generally not feasible to verify calculated deposition estimates and so most dry deposition estimates calculated from ambient air concentrations are usually presented without verification. Spatial estimates of dry deposition should be regarded as extremely preliminary (Hunter/ESE, 1989).

The current Clean Air Status and Trends Network (CASTNET), which incorporates the National Dry Deposition Network (NDDN), recommends against extending deposition velocity and dry deposition values determined at a particular site to a regional basis (Dr. Ralph Baumgardner, CASTNET Project Officer, personal communication). Deposition velocities, in addition, are not well defined for deposition to water surfaces. The network measures  $\text{NH}_4^+$ ,  $\text{HNO}_3$ , and  $\text{NO}_3^-$  in the particulate phase. The deposition velocities of  $\text{NH}_3$ ,  $\text{NO}$ , and  $\text{NO}_2$  are considered small due to the gaseous state and those species are not monitored. No trace metal or phosphorus data are available.

One of the CASTNET sites is in Sumatra, in the Florida panhandle region. In the most recent data available for this site (Table 16), total  $\text{NO}_3^-$  depositions, range between 1.4 and 1.9 times wet deposition (Mr. Ralph Baumgardner, CASTNET Project Officer, personal communication), although the annual total remained relatively constant between the two years.

Table 16. Wet, dry, and total deposition at the NDDN Sumatra, Florida site (Dr. Ralph Baumgardner, personal communication).

Year	Wet Deposition	Dry Deposition	Total Deposition
	kg N/ha/yr	kg N/ha/yr	kg N/ha/yr
1990	1.56	1.45	3.01
1991	2.49	1.06	3.55

Dry deposition in the FADS (Hunter/ESE, 1989) study was calculated both from dry bucket measurements and from ambient air concentrations and published deposition velocities. For  $\text{NO}_x$ , dry bucket deposition was lower by factor of 5, in comparison to the ambient air method of calculation, so sample contamination from birds and bugs was not an apparent problem as has been the case elsewhere. For  $\text{NO}_3^-$  and  $\text{NH}_4^+$ , dry bucket depositions were 0.55 and 1.14 kg N/ha/year, respectively, for the 1981- 1982 sampling year. Calculated deposition was 4.00 kg N/ha/year (26 % as  $\text{HNO}_3$ , the remainder as  $\text{NO}_2$ ), for the period 1982-1983. Any changes in ambient air concentrations between the two times periods is not available. Velocities used were 1.5 cm/sec for  $\text{HNO}_3$  and 0.3 cm/sec for  $\text{NO}_2$ . Using the calculation method, total deposition is 2.88 times the measured wet deposition during 1982-1983 at the Zephyrhills site. For the other 6 sites active at this time, the factor of total deposition: wet deposition ranged from 2.13 to 2.97.

The FADS and NADP data have been further reduced by Baker (1991) to determine regional dry deposition values. Dry bucket data was assumed to be representative of the deposition of larger particulates and was combined with the computed deposition used for the  $\text{HNO}_3$  and  $\text{NO}_2$  components. Velocities used were 1.0 cm/sec for  $\text{HNO}_3$  and 0.1 cm/sec for  $\text{NO}_2$ . For dry bucket  $\text{NH}_4^+$  and  $\text{NO}_3^-$  in Florida, total deposition was 1.39 and 1.19 times the wet deposition, i.e. the bulk of these ions was delivered in rainfall rather than dry deposition. Combining the dry bucket data with the ambient air data, total deposition at the Tampa site is estimated to be 2.16 times the measured wet deposition. Uncertainties in the estimate may be on the order of 15-20 % or greater if uncertainties in  $V_d$  are incorporated. The total: wet deposition ratio of 2.16 is used in the determination of current and benchmark loads to the Tampa Bay system (Zarbock, 1993, 1994) and is strictly applicable for inorganic nitrogen data only.

Edgerton and Lavery (1990) have also examined FADS, NADP, and NDDN data. From a range of seasonal deposition velocities (0.06 - 0.20 cm/sec for  $\text{NO}_3^-$  and 1.5 -2.0 cm/sec for  $\text{HNO}_3$ ),  $\text{NO}_3^-$  dry deposition appears to be between 1.40 and 1.12 kg/ha/year for the central Florida region with ratios of total: wet deposition between 1.66 and 1.71.

Fanning (1992) recognized the general omission of organic nitrogen in the summary of pertinent wet, dry, and total deposition values for nitrogen to Tampa Bay. As the larger particles form a relatively small percent of dry deposition, organic nitrogen in dry deposition is not expected to be substantial. Data presented by NURP (Noel, *et al.*, 1987), Rushton( 1993), CCI (1992), however, indicate that organic nitrogen could represent between 30 % and 60 % of the wet deposition, and therefore 15 % - 30% of the total nitrogenous deposition (using the total: wet deposition ration of 2.16).

In the CBAD study (Baker, *et al.*, 1992), dry deposition was estimated for number of metals from deposition velocities and ambient concentrations. Deposition velocities ranged between 1.4 and 4.0 cm/sec for crustal elements (assumed to be associated with larger soil particles) and between 0.26 and 0.72 cm/sec for non-crustal materials (aerosols from high temperature combustion). Because of the distribution of crustal to non-crustal elements, and the range in velocities, the estimate of dry deposition has an uncertainty of +/- 50 % and values are listed in Table 15, above.

## Transport from the Watershed

Atmospheric deposition, when applied to the areal surface of a waterbody, is a direct and immediate impact. Materials deposited to the watershed, however, if not subject to immediate runoff, become involved in biogenic cycling, physical removal through sedimentation, plant and biota uptake, and the like. Recycling will also liberate materials to surface waters such that they can be transported downstream. The degree to which materials are retained by a watershed is also a function of chemical reactivity with soils. In the absence of anthropogenic “imports” into the watershed, flux measurements of both atmospheric deposition and fluvial runoff will indicate the proportion of atmospheric loading to the watershed (indirect atmospheric deposition) which eventually is transported to the Bay.

Direct measurements of fluvial loads are unavailable for much of the Tampa Bay watershed. Some of the older surface water quality data on trace metals are felt to be unreliable for the sensitivity of analyses attempted. Gaged and monitored stations are also usually upstream some distance on the various tributaries, and do not include the water quantities or loadings generated by the downstream portions representing the bulk of the urbanized area of the watershed. Several projects are underway to address tributary loading the Bay, but in the interim, calculated non-point source loads will be presented as surrogates for true fluvial loads.

Simplistically, the non-point source loads generated by a watershed are calculated from size, empirically determined land use loading rates, and the proportion of various land uses within the watershed. A comparison, therefore, of the watershed non-point source loadings with the estimated atmospheric deposition to a watershed will allow an evaluation of the proportion of atmospheric deposition which may be represented in runoff concentrations. This proportion will overestimate the eventual atmospheric contributions to the Bay to some unknown extent, as watershed and in-stream removal processes have not been applied.

The proportion of indirect atmospheric loading which eventually reaches an estuary is to some extent a function of the ratio of watershed area to open water surface (drainage ratio). For elements which are not retained 100% by the watershed, increasing drainage ratio results in increasing element burdens downstream (Blais and Kalff, 1993). For lakes in southern Canada, lead, zinc, and to a lesser extent, copper, were almost completely retained in the watershed (Blais and Kalff, 1993). For Tampa Bay, the drainage ratio is approximately 6, relatively small in comparison to Chesapeake (14.5) and Delaware Bays (17) (Scudlark and Church, 1993). Implications are that only a small fraction of trace metals would be exported to the Bay, except for the noteworthy point that much of the impervious area in the watershed is directly adjacent to the Bay, and that runoff has little time to equilibrate with soils. For mercury, on the other hand, retention efficiency is reported near 75 % (Swain, *et al.*, 1992), such that 25 % of deposited mercury is exported from the watershed.

Retention rates of nutrient species also vary with land use and values range between 25 % and 98% (Scudlark and Church, 1993). For Delaware Bay, Scudlark and Church (1993) assumed that 80% of dissolved inorganic nitrogen species (DIN) were retained in the watershed, and that further, 50% of the remaining DIN was removed in-stream through phytoplankton uptake and denitrification. In the Chesapeake Bay, Tyler (1988) employed a range of nitrate retention rates

of between 100% and 76 % for such land uses as forests and croplands and applied further delivery ratios of 50 % to 80 % to account for in-stream removals. Uncertainties in the retention values and delivery ratios are quite high, but nitrate loading to the Chesapeake which is attributable to atmospheric deposition is on the order of 25 %. Hinga, *et al.*, (1991) revises the Chesapeake Bay estimate upwards to near 32%) and for Ocklockonee Bay in north Florida, estimates that 100 % of the nitrogen loading was from atmospheric sources.

In order to estimate the potential magnitude of the impacts of atmospheric deposition on Tampa Bay, the range of deposition values as described earlier in this document were computed and compared to some recent non-point source loadings estimates. Not included in either the non-point source estimation or the atmospheric component are any in-stream processes. Non-point source loadings lower than the calculated atmospheric deposition could represent some immediate watershed retention which may take place in the small basins typically sampled for land use-specific loading rates. Actual non-point source loads to the Bay, whether calculated from non-point source loading or from deposition estimates, would be expected to be some factor lower due to in-stream removals. The estimates used for in-stream removals, however, would be the same in either case.

Initial estimates of non-point source loadings to Tampa Bay were provided by Dames & Moore (1990) with an early version of the NonPoint Source Load Analysis Model (NPSLAM), using land use, land use specific runoff and loading coefficients, soil types and rainfall. From the 1990 land uses and recommended mass loading rates by land use (summarized from a number of empirically determined values), watershed average loading rates were calculated (Table 17). The rates do not account for instream assimilation or other removal processes that may influence water quality before discharge to the Bay, although removal efficiencies were assigned to stormwater treatment facilities.

Table 17. Average watershed non-point source mass loading rates (from data in Dames & Moore, 1990).

Chemical Species	Average Watershed Non-Point Source Loading Rates
kg/ha/yr	
Total N	8.56
PO <sub>4</sub> -P	0.76
Total P	1.13
Zinc	0.23
Lead	0.31

NPSLAM reportedly overpredicted runoff in some instances and subsequent modeling efforts included the addition of seasonal variations in non-linear rainfall:runoff relationships, short term

antecedent moisture conditions, and lagged rainfall effects (Zarbock, *et al.*, 1993). Where gaged basins were accompanied by a water quality record, loadings were computed directly. For ungaged basins, the model utilized the rainfall:runoff simulations and land use loading factors. Point sources, spring discharges, groundwater flows, fugitive emissions, and atmospheric deposition were also incorporated in determining the total loads of nutrient to Tampa Bay.

For the various basins surrounding the Bay, Zarbock, *et al.*, (1993) computed non-point source loadings of nitrogen and phosphorus, of approximately 2,470 metric tons/year and 626 metric ton/year, respectively, during a 1985-1991 time period. Urban and agricultural land uses accounted for the bulk of these loadings. As for NPSLAM, no instream processes were simulated. The non-point source loading comprised some 51% of the total nitrogen load to the Bay and 16% of the total phosphorus load. Atmospheric deposition direct to the Bay is estimated to consist of 28% and 8% of the total nitrogen and phosphorus load to the Bay, respectively, but this figure is to be revised upwards in the near future (Zarbock, 1994) to account for higher dry deposition loadings. A total watershed area of 5,571 km<sup>2</sup> (5,895 km<sup>2</sup> less the internally drained portions), was then used to compute the average watershed non-point source loading rates below (Table 18). Phosphorus loadings were almost identical, but nitrogen loadings are approximately one-half those calculated previously (Dames & Moore, 1990) (Table 19), although the range of land use runoff concentrations employed in each work appears similar. Differences in land use percentages, annual rainfall, and runoff oversimulation observed for some basins with NPSLAM may account for the discrepancy.

Table 18. Average watershed non-point source mass loading rates (from data in Zarbock, *et al.*, 1993).

Chemical Species	Average Watershed Non-Point Source Loading Rates
	kg/ha/yr
Total N	4.44
Total P	1.06

The estimated non-point source loading rates were then compared to the various atmospheric deposition values previously described. Where data exist only as precipitation concentrations, 135.7 cm (53.44 in) was used as an average annual rainfall to calculate wet deposition rates. The rainfall value is the arithmetic mean of all individual basin rainfalls, as interpolated by Dames & Moore (1990). If nutrient or metal deposition rates are available under an urban or rural classification, the land uses tabulated in Dames & Moore (1990) were used to allocate the watershed into 33 % “urban” (commercial, residential, industrial, and mining) and 67 % “rural” for first order estimates.

Table 19. Land use apportionment for the Tampa Bay watershed (Dames & Moore, 1990).

Land Use	Percentage (%)
Industrial	1
Wetland /open water	3
Commercial	6
Mining	8
Low-density, single family, or multi-family residential	18
Recreational /open areas	24
Agricultural	40

For phosphorus, where both total and phosphate atmospheric deposition rates are available, phosphate comprises approximately 50% of the total (Rushton, 1991; Noel, *et al.*, 1987). Phosphate values, in kg/ha/year, were 0.15 for the rural Carey Forest (Allen and Sutton, 1990), 0.30 for Tampa (Rushton, 1991), 0.58 for the Zephyrhills FADS site (ESE, 1986), and 1.22 for the NURP data (Noel, *et al.*, 1987). The FADS site was felt to be influenced by mining nearby, and the NURP data appear excessive, and so a value between 0.60 and 1.16 kg/ha/year of phosphorus, is likely appropriate for the watershed. These values are for wet deposition only, but compare quite well with the 1.06 kg/ha/year determined by Zarbock, *et al.*, (1993) for non-point source loadings.

Nitrogen data are more complex, due the variety of species sampled and estimated. Wet deposition for  $\text{NO}_3^-$  in Florida converge on 2.46 kg/ha/year with Rushton (1993), the FADS Zephyrhills site (ESE, 1986), the Carey Forest (Allen and Sutton, 1990), Cross Bayou (PCDEM, 1993c) at 2.61, 2.05, 2.48, and 2.71 kg/ha/year, respectively. Again the NURP values for  $\text{NO}_3^-$  are quite high, 4.16 kg/ha/year, and were not included in the estimate. Wet deposition for  $\text{NH}_4^+$  averages 1.66 kg/ha/year, as does the most reliable of the organic nitrogen wet deposition rates (Rushton, 1993). The dry deposition rates for nitrogen species can be substantial. Total nitrogen deposition rates are presented by a number of authors, but generally lack quantification of at least one of the nitrogen species. Organic nitrogen and  $\text{NH}_4^+$  wet deposition loading from Rushton (1993), together with the factor (1.16) for dry deposition reported by Baker (1993), were used to “complete” the analytical suites.

Total nitrogen deposition (wet plus dry) at the Carey Forest was estimated at 9.92 kg/ha/year, 7.79 kg/ha/year at the FADS Zephyrhills site, and 12.6 kg/ha/year for the Rushton (1993) sites. NURP nitrogen data were much higher, with extremely high concentrations of organic nitrogen and estimates of total deposition of greater than 22 kg/ha/year. Values of watershed non-point source loadings computed from Zarbock, *et al.*, (1993) were near 4.44 kg/ha/year. Wet deposition loadings alone from Rushton (1993) are 5.85 kg/ha/year and Brezonik, *et al.*, (1981) presents a range of 5.8 to 7.6 kg/ha/year.



With estimated atmospheric deposition so much larger than modeled non-point source loadings, it would seem to indicate that a substantial fraction of atmospheric deposition is retained immediately by the watershed and that all of the nitrogen present in runoff has an atmospheric origin. Comparison of computed non-point source loads to the more recent deposition estimates indicate that up to 45% of the atmospheric load may be retained; comparison with the older atmospheric estimates (from NURP) indicate that 80% of atmospheric nitrogen is retained on the watershed. The 80% figure agrees well with estimates from other regions (Scudlark and Church, 1993) and may be more appropriate as many of the land use runoff coefficients and loadings developed for estimating non-point source loads were determined during the time that NURP data were being collected.

Atmospheric loading rates for metals, with the exception of the NURP data, were not specific to the Tampa Bay area or even Florida. Wet deposition loadings determined by a number of researchers were very comparable to the total of wet and dry determinations elsewhere. As the data were not specific for Florida, the mean of the reported values for total depositions was used as a working value. With the exception of zinc, urban and rural values by Nriagu (1992) generally bracketed the average total deposition, while again NURP data appeared as a substantial outlier (Table 20).

Table 20. Summary of trace metal atmospheric deposition rates.

Trace Metal	Rural <sup>a</sup>	Average Total Deposition <sup>b</sup>	Urban <sup>a</sup>	NURP
	g/ha/yr	g/ha/yr	g/ha/yr	g/ha/yr
Cadmium	0.68	1.8	4.8	11
Copper	6.1	14	38.0	489
Lead	19.0	26	81.0	81
Zinc	11.0	60	48.0	1680
Mercury	0.34 <sup>c</sup>	0.026	0.88 <sup>c</sup>	

<sup>a</sup> Nriagu, 1992

<sup>b</sup> Average of data from Vermette *et al.* (1992), GLAD, Orr *et al.* (1990), and Church and Scudlark (1990).

<sup>c</sup> Older data, potentially unreliable for mercury concentrations.

Using the 5,571 and 958 km<sup>2</sup> values for the watershed and Bay area, total loads to the Bay were computed (Table 21).

Table 21. Estimates of annual trace metal atmospheric deposition to Tampa Bay and the surrounding watershed.

Trace Metal	Atmospheric Deposition	----- Total Deposition -----	
	g/ha/yr	to Bay kg/yr	to Watershed kg/yr
Cadmium	1.58	151	880
Copper	12.6	1,210	7,020
Lead	26.7	2,260	14,900
Zinc	54.1	5,180	30,100
Mercury	0.026	2.5	14.5

Of those parameters with estimates, zinc and lead average watershed loading rates were used to compute non-point source loads of 128,000 kg/year of zinc and 173,000 kg/year of lead. The non-point source values are a factor of 4 and a factor of 12, respectively, greater than that estimated from the atmospheric deposition rates. The atmospheric deposition rates were from the northeast, and should, if anything, be substantially greater than Florida values. Review of the references used to determine land use loading factors in Dames & Moore (1990), however, reveal that many date from the early 1980's. The NURP data from this time period are consistently elevated in many parameters, including lead and zinc. Examination of national trends in lead emissions show a decrease by a factor of 12.3 between 1981 and 1991 (USEPA, 1993). Emissions decreases are primarily in transportation, so the Tampa area should have experienced comparable improvements. Reducing the non-point source 173,000 kg Pb/year by the factor of 12.3 obtains a lead deposition of 14,100 kg/year, comparing most favorably with the atmospheric estimate of 14,900 kg/year.

No national emissions data for zinc are available, but it is apparent that NURP rainfall zinc concentrations are up to 30 times higher than data collected more recently. Aside from the possible site-to-site differences, there is a strong potential that the reduced depositions reflect a real improvement in air quality. One is urged to the conclusion that NURP data were representative of the time they were collected, but that improvements in ambient air quality have made the NURP data, and land use loading rates determined concurrently, obsolete for at least some parameters.

## Summary

A compilation of available literature and information pertinent to Tampa Bay atmospheric deposition. Sources, emissions, available ambient air quality data, precipitation quality, reported ranges of wet and dry deposition rates, previous and ongoing monitoring programs were summarized. Nitrogen and phosphorus were emphasized, (except for deposition rates) as sources and emissions of trace metals were to be addressed in other documents.

Anthropogenic sources of nitrogen oxides within the Tampa Bay watershed are substantial, and stationary sources alone comprise 25% of the State's total NO<sub>x</sub> emissions. Stationary sources are dominated by two utility stations on the shores of Tampa Bay. Mobile sources are also noteworthy, forming some 30% of the total NO<sub>x</sub> emissions for the nine county area surrounding the Bay. Sources of phosphorus are relatively unstudied and unquantified. Estimates of fugitive emissions from the transportation of phosphatic materials have been made based on shipping tonnage.

Ambient air quality data were reviewed and indicate that no NAAQS (for NO<sub>2</sub> and Pb) were violated in 1992. Stations are concentrated in areas of expected poor air quality (urban) or near known point sources, and so do not provide data for the bulk of the watershed.

Precipitation data show distinct seasonal patterns, with the majority of the loading occurring during the summer wet season. Power usage patterns, meteorological circulation patterns, or greater efficiency of rainfall scavenging of materials with convective storms could be responsible. Comparisons of data sets for the Tampa Bay area show distinctly elevated urban concentrations and depositions in rainfall. The most urban data set, however, is older and could reflect true changes in ambient air quality and subsequent deposition. Organic nitrogen and phosphorus components in deposition are comparatively poorly characterized and could be substantial.

Estimates of the atmospheric deposition of total nitrogen range between 7.8 and 22 kg/ha/yr, while non-point source loadings are modeled with an average for the watershed of 4.4 kg/ha/yr. A net retention within the watershed of between 50 % and 80% of the atmospheric nitrogen is implied. These figures are in general agreement with literature from other areas of the United States. Estimates of atmospheric deposition of phosphorus (0.6 - 1.16 kg/ha/yr) compare favorably with estimates of non-point source loading (1.06 kg/ha/yr) of this element. No substantive net retention of phosphorus within individual basins is observed, with a potential for anthropogenic enrichment.

For deposition of trace metals, data from many differing time periods and regions was presented. There are little data on trace metal deposition for the region, and data from the mid-Atlantic and north central states were used for first order estimates. Atmospheric loadings of trace metals were computed, but where non-point source calculations were available (for lead and zinc), were substantially lower than the loadings expected from the land uses within the Tampa Bay watershed. The reduced loadings of lead can be directly related to reduction in lead emissions, and the potential exists that land use loading coefficients determined in earlier periods do not accurately reflect current conditions.

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## Appendix A

### Estimated emissions of NO<sub>x</sub> by county and source

(Units are tons/year)

#### HILLSBOROUGH COUNTY:

	<u>NOX</u>
Alumax Extrusions	14. 4900
Anheuser-Busch, Inc.	28. 0100
Ball Packaging Product Group	6. 3126
Cargill Fertilizer, Inc.	21. 2030
CF Industries, Inc., Plant City Phosp	40. 3940
Coronet Industries, Inc.	78. 0650
Florida Steel Corp.	36. 5700
Hillsborough Cty. Resource Recovery Fac.	770. 5920
IMC-Agrico Co. (Port Sutton)	94. 4600
Master Packaging, Inc.	5. 8070
Mobil Mining and Minerals Company	75. 7200
National Gypsum Company	62. 6600
Nitram, Inc.	24. 4460
Tampa City McKay Bay Refuse-To-Energy	513. 2000
Teco-Big Bend Sta.	50, 873. 7600
Teco-Gannon Sta.	38, 669. 3000
Taco-Hookers Pt. Sta.	629. 3100

Totals for Hillsborough County:

91,951.8346

#### MANATEE COUNTY:

	<u>NOX</u>
Ajax Paving Industries, Inc.	16. 1100
Coastal Fuels Marketing, Inc.	10.8300
Florida Power & Light	6, 437. 0000
Tropicana Products, Inc.	720. 0247

Totals for Manatee County:

7,197.3335

#### PINELLAS COUNTY:

	<u>NOX</u>
Florida Power	591.0610
Florida Power	141. 9550
Florida Power Corporation	3, 847. 5870
Pinellas Co. Resource Recovery Facility	1, 253. 7000

Totals for Pinellas County:

5,839.0520

(Units are tons/year)

<u>SARASOTA COUNTY:</u>	<u>NOX</u>	
Apac-Florida, Inc.	8.5100	
Gator Asphalt Company	6.8900	
Totals for Sarasota County:		15.4000
<u>CITRUS COUNTY:</u>	<u>NOX</u>	
Florida Power	<b>48,373.4260</b>	
Totals for Citrus County:		<b>48,373.4260</b>
<u>HARDEE COUNTY:</u>	<u>NOX</u>	
Teco Power Services Corp.	<b>0.000</b>	
Totals for Hardee County:		<b>0.000</b>
<u>HERNANDO COUNTY:</u>	<u>NOX</u>	
Central Power & Lime, Inc.	1,097.1200	
Florida Mining & Materials	330.0301	
Totals for Hernando County:		<b>1,429.0874</b>
<u>PASCO COUNTY:</u>	<u>NOX</u>	
Couch Const. Co.	13.7200	
Florida Power	5,116.7410	
Lykes Pasco, Inc.	141.7561	
Overstreet Paving Co.	23.0700	
Pasco County Resource Recovery	908.9000	
R.E. Purcell Const. Co.	14.7902	
Totals for Pasco County:		<b>6,218.9773</b>

(Units are tons/year)

POLK COUNTY:

NOX

Aluminum Company of America	12.8700
Bio-Medical Service Corp. of Ga. (BFI)	15.2000
Cargill Citro-America, Inc.	40.0890
Cargill Fertilizer, Inc.	121.2000
Citrus World, Inc.	75.9300
City of Lakeland - Larsen Power Station	256.5301
City of Lakeland - McIntosh Power Station	3,536.0700
Coca Cola	116.2498
Farmland Industries, Inc.	581.9060
Florida Juice, Inc.	82.5101
IMC Fertilizer, Inc. - Prairie	19.6686
IMC-Agrico Company (Kingsfor Mine)	15.0450
IMC-Agrico Chemical Company (New Wales)	349.9271
IMC-Agrico Company	54.1001
IMC-Agrico Company (Nichols Plant)	22.8900
IMC-Agrico Company (Noralyne Mine)	21.0251
Lakeland Drum	5.3400
Mobil Mining & Minerals Company	8.3100
Mulberry Phosphates, Inc.	7.3190
Orange-Co of Florida, Inc.	14.6654
Owens-Brockway Glass Container Inc.	325.8000
SFE Citrus Processors, L. P., Ltd.	12.4401
Standard Sand & Silica Co.	9.9546
Sun Pat Foods, Inc.	6.5490
U. S. Agri-Chemicals Corporation	48.0000

Totals for Polk County:

**5, 778. 2512**

CHARLOTTE COUNTY

NOX

Ajax Paving Industries	<b>16. 3200</b>
Apac-Florida (Macasphalt)	<b>11. 2000</b>

Totals for Charlotte County:

27.9000